



# In-situ synthesis of CaO and SiO<sub>2</sub> nanoparticles onto jute fabrics: exploring the multifunctionality

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**Abstract** Fibrous based materials with exceptional functionalities have become a target of interest for researchers worldwide. One promising strategy for the development of active fibrous structures with improved functions is their functionalization with nanoparticles (NPs). In this work, jute fabrics were functionalized with calcium oxide (CaO) and silica (SiO<sub>2</sub>) NPs in order to obtain fibrous structures with several properties including, hydrophobicity, UV protection, antibacterial activity and the degradation of dyes ability. CaO and SiO<sub>2</sub> NPs were synthesized via a simple in situ method, using sodium hydroxide (NaOH) as the reducing agent and water as solvent. Parallely, the in situ synthesis of the CaO NPs was also tested. The final systems were fully characterized using Attenuated Total Reflectance-Fourier Transform Infrared Spectroscopy, Ground-State Diffuse Reflectance and Field Emission Scanning Electron Microscopy and Energy Dispersive Spectroscopy. All

these techniques confirmed the successful synthesis of the NPs as well as their well distributed presence onto the fabrics. The samples exhibited very good contact angle values, reaching 143.7° for the fabrics functionalized with CaO and SiO<sub>2</sub> NPs and great values of ultraviolet (UV) protection factors (UPF), reaching 50+ for the functionalized fabrics. The developed systems also exhibited antibacterial activity against *Staphylococcus aureus* and *Escherichia coli* and photocatalytic activity for the degradation of methylene blue. The wash durability of the nanocoating was also evaluated, confirming that the SiO<sub>2</sub> improved the NPs' anchorage onto the fabrics. Overall, this work presents for the first time, the development of jute fabrics with CaO–SiO<sub>2</sub> NPs with several functionalities.

**Keywords** Jute fibers · In situ-synthesis · Nanoparticles · Antibacterial activity · Hydrophobicity · Photodegradation

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## Introduction

The development of multifunctional fibrous based materials has become a target of interest for many researchers all over the world. Nanotechnology could be responsible for the enhancement of textiles properties, since the incorporation of nanomaterials can

introduce several functionalities, that can include, easy/self-cleaning, decomposition of chemical agents, antimicrobial activity, flame retardancy, UV-protection, antistatic properties and monitoring/sensing behavior (Costa et al. 2018; Joshi and Adak 2018; P Ferreira et al. 2018). The functionalization of fibers with NPs is an eminent alternative in the development of functional fibrous systems.

One of the biggest concerns for governments worldwide has been the protection against emerging infectious diseases. Antibacterial textile substrates prevent the growth, or even kill microorganisms, reducing the transmission of these diseases (Qiu et al. 2020). Thus, the functionalization of fibers in order to obtain antibacterial activity is essential. Hydrophobic textiles have also attracted some attention due to their capacity for liquid repellency, self-cleaning and unidirectional fluid transport, which can turn these materials into great alternatives to be used for several applications (Tissera et al. 2015).

The excessive unprotected solar exposure can lead to several health issues, mostly skin damage, like aging, acne and even melanoma and carcinoma. An effective UV protection material, that prevents the harmful effects caused by the sun, is of huge importance (Sasani Ghamsari et al. 2017).

The disposal of several noxious pollutants without a proper treatment has been perpetually increasing. This is one of the biggest causes of environmental pollution. Therefore, there is the need to develop various methods for the removal of these pollutants. One of the methods with this finality that has been showing the greatest potential is photocatalysis, especially for the removal of organic pollutants. Semiconductors have been reported as one of the best materials for photocatalysis, since they are able to degrade organic compounds using irradiation of light with different wavelengths (Nagpal and Kakkar 2019).

Dyes are considered the most common industrial pollutant and their concentrations can be easily monitored by a UV–Vis spectrometer, which makes them one of the most used compounds on photocatalytic degradation tests. MB is a poisonous dye that causes breathing hazards, vomiting and mental disorders (Balu et al. 2018). Due to the noxious effects that dyes have on humans, the development of catalytically active fabrics, with self-decontaminating properties, present great potential for application in chemical

protective clothing against several harmful chemicals (Boufi et al. 2019).

NPs present great potential for imparting various functionalities to textiles due to their nanoscale dimension and very high surface area to volume ratio (Mary Ealias and Saravanakumar 2017). One type of NPs that has obtained considerable research attention is metal oxide NPs, due to their exceptional electrical, optical, magnetic and catalytic properties (Chavali and Nikolova 2019). This excellent set of characteristics have allowed their wide application in several areas, including agriculture, healthcare and industry (Nguyen et al. 2018).

Metal oxide NPs, namely: zinc oxide (ZnO), titanium oxide (TiO<sub>2</sub>), SiO<sub>2</sub>, aluminum oxide (Al<sub>2</sub>O<sub>3</sub>), magnesium oxide (MgO) and cerium oxide (CeO<sub>2</sub>) can be used for UV-protection purposes and decomposition of organic compounds, pesticides and microorganisms (Becheri et al. 2008; Duan et al. 2011). Within these particles, CaO NPs have shown a great potential as a cheap antibacterial agent, a photocatalyst for dye degradation and nanosized semiconductor. CaO NPs are excellent candidates for fibrous materials functionalization because when compared with other metal oxides, CaO is considered as a safe material to human beings, biocompatible, non-toxic and acceptable as food additive. It also has excellent antimicrobial potential, is adsorbent and in particular is known as a destructive adsorbent for toxic chemical agents. It is important to take all these described properties into account when thinking in textile applications since the developed materials could be in close contact with human skin (Bharathiraja et al. 2018). At the same time, SiO<sub>2</sub> NPs due to their easy and low-cost large-scale preparation, good biocompatibility and large specific area are widely used for several applications. One of the most important characteristics of SiO<sub>2</sub> NPs is the possibility to merge with other materials, to combine their functionalities (Peerzada et al. 2019). The addition of SiO<sub>2</sub> can improve the CaO NPs properties, due to its high active surface (Gunathilake and Jaroniec 2016).

Nowadays, with the environmental growing consciousness, there has been an increasing interest in the use of natural fibers instead of the synthetic ones, because of their biodegradability, biocompatibility, low-weight, high abundance and low-cost, which makes this kind of fibers an exceptional material for the development of an environmentally-friendly

fibrous system (Costa et al. 2018). Jute is the second most produced fiber worldwide, after cotton. In addition to its great abundance in nature, jute is also cheap, biodegradable and has a low density. It also presents the possibility of being woven into different shapes and forms (Ivanovska et al. 2019; P Ferreira et al. 2018). These types of fibers also present very good mechanical properties, and the cellulose matrix of jute fabrics has already been reported has a great candidate for the dispersion of metal oxide NPs (Sahu et al. 2017). Jute also presents a lignin content of approximately 16%, which is very high when compared to other bast fibers (del Río et al. 2009). In addition, can be easily manipulated and can be applied in several situations, especially for rapid environmental remediation (Hao et al. 2015).

Apart from the use of sustainable materials, like the natural fibers, it is also essential to reduce the environmental impact of the various industrial processes used for fibers' functionalization. In several research works, CaO NPs were synthesized using very high calcination temperatures, between 500 and 700 °C (Alavi and Morsali 2010; Anantharaman et al. 2016; Bharathiraja et al. 2018). Thus, avoiding the use of very high calcination temperatures is of major importance.

In this work, jute fabrics were functionalized with CaO and SiO<sub>2</sub> NPs and several functionalities were studied. In a first stage, the pre-treated fabrics were functionalized with CaO and SiO<sub>2</sub> NPs by a very simple in situ synthesis. The in situ synthesis of only the CaO NPs was also tested using as calcination temperature 150 °C. All the developed fibrous systems were characterized by ATR-FTIR, GSDR and FESEM-EDS, that confirmed the successful synthesis of the two types of NPs. Functional properties of the samples were also evaluated, namely hydrophobicity by the contact angle measurement, UV protection behaviour, MB degradation and antibacterial tests against *Staphylococcus aureus* and *Escherichia coli*.

## Experimental

### Materials

The jute fabrics used in this work were produced using 100% natural jute yarns (Bangladesh origin) with weight of 430 g/m<sup>2</sup>, supplied from RCS (Braga,

Portugal). Calcium nitrate tetrahydrate (Ca(NO<sub>3</sub>)<sub>2</sub>·4H<sub>2</sub>O) was purchase from Scharlau and NaOH from Normax Chem. Tetraethyl orthosilicate (TEOS) was obtained from Sigma-Aldrich. The ammonia (NH<sub>3</sub>) was acquired from AnalaR NORMAPUR. The ethanol absolute used was purchased at José Manuel Gomes dos Santos. A Memmert UNE 800 oven was used. UV blacklight lamp (15 W) from HQ<sup>TM</sup> was used for photocatalytic tests.

### Fabrics pre-treatment

Jute fabrics (5 cm × 5 cm) were washed with 5% (v/v) of non-ionic detergent at 80 °C for 30 min to remove the adhered impurities (waxes, fats, etc.). The fabrics were further cleaned in distilled water at 70 °C for 30 min and dried at 100 °C for 1 h.

### Functionalization of the fabrics with CaO and SiO<sub>2</sub> nanoparticles

#### *In-situ synthesis of CaO and CaO–SiO<sub>2</sub> onto the fabrics*

For the in situ synthesis of only the CaO NPs onto the jute fabrics, 10 ml of a NaOH aqueous solution (1 M) were added drop by drop (1 ml/min) in 50 mL of a Ca(NO<sub>3</sub>)<sub>2</sub> aqueous solution (0.2 M), under mechanical stirring at 50 °C for 1 h. Then, the samples were removed from the solution and heated at 150 °C for 6 h. To perform the in situ synthesis of the CaO and SiO<sub>2</sub> NPs the process was the same as the one referred before, until the end of the 1-hour time reaction. After the reaction mixture reached room temperature, 100 ml of ethanol and 4 mL of NH<sub>3</sub> were added. Then, 5 mL of a TEOS ethanolic solution (50% vol TEOS in ethanol) were added dropwise, and the mixture was kept under constant stirring, at room temperature for 24 h. At the end, the fabric samples were removed from the solution, and heated at 150 °C for 6 h.

### Sample characterization

#### *Attenuated Total Reflectance-Fourier Transform Infrared Spectroscopy (ATR-FTIR)*

The chemical composition of the developed systems was studied by ATR-FTIR analysis using an

IRAffinity-1S, SHIMADZU equipment (Kyoto, Japan). The spectra were obtained in transmittance mode, in the range of 400–4000  $\text{cm}^{-1}$ . The samples were analysed 3 times in different places, in order to ensure the analysis' homogeneity.

#### *Ground-State Diffuse Reflectance (GSDR)*

The fabrics functionalized with the two types of NPs were analysed by GSDR. The transmittance and reflectance spectra were recorded using a Spectrophotometer UV-2600 (Schimadzu) with the ISR\_2600 Plus detector. The equipment was calibrated using as blank barium sulphate (full reflectance).

Different places of the samples were analysed, in order to ensure the analysis' homogeneity. The reflectance spectra were recorded, in the 200 to 700 nm wavelength, and the remission function was calculated accordingly with the Kubelka–Munk equation:

$$\frac{K}{S} = \frac{[(1 - R)]^2}{2 \times R}$$

where K represents the absorption coefficient, S the dispersion coefficient and R the reflectance (Yang and Kruse 2004).

#### *Field Emission Scanning Electron Microscopy (FESEM) and Energy Dispersive Spectroscopy (EDS)*

The surface of the fabrics functionalized with the NPs was analysed by FESEM using NOVA 200 Nano SEM from FEI Company (Hillsboro, OR, USA). The samples were coated with a thin film (20 nm) of Gold (Au) and Palladium (Pd) before the analysis. The images were taken in topographic mode with an accelerated voltage of 10 kV. EDS technique (Hillsboro, OR, USA) coupled to FESEM was also performed, in order to evaluate the elemental composition of the samples, using EDAX Si(Li) detector with 15 kV of acceleration voltage.

#### *UV-visible absorption spectra*

The UV-visible absorption spectra of the MB solutions were obtained using a Unicam double beam scanning UV-visible spectrophotometer

(ThermoFisher, Waltham, MA, USA) in the spectral range from 500 to 750 nm at room temperature. Several aliquots were collected from the solutions in study and diluted to prevent deviation from Beer's Law linearity.

#### *Thermogravimetric analysis (TGA)*

TGA analysis was performed using a STA 700 SCANSCI in order to do an estimation of the amount of NPs incorporated onto the jute fabrics. Jute fabrics without any functionalization and functionalized with CaO and CaO–SiO<sub>2</sub> NPs were subjected to a heating process from room temperature to 600 °C under nitrogen flow and with a heating rate of 20 °C/min. The percentage of NPs was calculated using the following system of equations (Costa et al. 2018):

$$\begin{cases} \text{NPs \%} + \text{jute fabric \%} \left( \frac{\text{jute fabrics residues \%}}{100} \right) = \text{sample total residues \%} \\ \text{NPs \%} + \text{jute fabrics \%} = 100\% \end{cases}$$

Considering that the NPs have residues of 100% and the sample with NPs after thermal treatment is composed of NPs and jute fabric residue. Three replicates were analysed to ensure homogeneity.

#### *Functional properties evaluation*

##### *Water contact angle measurement*

In order to evaluate the hydrophobicity of the different functionalized fabric samples, the Water Contact Angle (WCA) was determined using a Contact Angle System (dataphysics) coupled to a high-resolution camera. To do the analysis, 5  $\mu\text{L}$  of distilled water were dispensed from the syringe onto the surface of the sample. Ten different places of each sample were measured, and the means and the standard deviations were calculated.

##### *UV protection*

In order to calculate the UPF and the UVA and UVB blocking values of the developed systems, the transmittance spectra of the samples were recorded in the 280–400 nm wavelength range, following EN13758-1, using a Spectrophotometer UV-2600 (Schimadzu) with the ISR\_2600 Plus detector. Different places of

the samples were analysed, in order to ensure the homogeneity of the analysis.

#### *Antibacterial activity tests*

The determination of antimicrobial activity was performed according to the internationally recognized Japanese industrial standard (JIS L 1902: 2002, ‘‘Testing method for antibacterial activity of textiles’’). The method used was the absorption method, which is a quantitative one. The microorganisms assayed were the Gram-positive bacteria *Staphylococcus aureus* (ATCC 6538) and the Gram-negative bacteria *Escherichia coli* (ATCC 434). Briefly, inocula of *E. coli* and *S. aureus* were prepared in  $20.0 \pm 0.1$  mL of TSB (Tryptic Soy Broth, Merck) and incubated for a period of 18 to 24 h at  $37 \pm 1$  °C under agitation (120 rpm). Samples were incubated for 24 h at  $37 \pm 1$  °C, without agitation. Then, 5 mL of physiological saline solution (8.5 g of NaCl and 2.0 g of non-ionic surfactant Tween 20 (Sigma Chemical Co.) per litre) were added to the samples, which were then vortexed (5 times, 5 s each time). The number of living cells was assessed by the serial dilution plate count method. All assays were performed in triplicate and repeated in three independent assays.

The percentage reduction of bacteria after 24 h incubation was calculated by the formula:

$$\% \text{Reduction} = [(B - A)/B] \times 100$$

where A = number of bacteria in the tested sample after 24 h, B = number of bacteria in the starting suspension before addition of the sample.

#### *Methylene blue degradation*

The photocatalytic activity of the CaO and SiO<sub>2</sub> NPs deposited on the fabrics was accessed by the degradation of MB as a model compound. The functionalized fabrics (0.5 g) were immersed in 25 mL of an aqueous solution of MB (10 mg/L) and irradiated by a UV blacklight lamp during 2, 5, 7 and 24 h. Aliquots of the MB solution (3 mL), in contact with the different fabric samples, were withdrawn and their absorbance was measured by UV–Visible spectroscopy after the 2, 5, 7 and 24 h’ time of UV irradiation, in order to monitor MB presence.

#### *Durability test: wash fastness*

In order to evaluate the wash durability of the nanocoating of CaO and CaO–SiO<sub>2</sub> onto the jute fabrics, jute, jute + CaO and jute + CaO–SiO<sub>2</sub> samples were placed in contact with distilled water and centrifugated at 200 rpm for 6 h, continuously. This procedure was adapted from the standard ISO6330-Textiles, Domestic Washing and drying, procedures for textile testing, in order to mimic the domestic washing programmes. According with this standard the agitation speed could range between 119 and 179 rpm, for delicate and durable press parameters, respectively. The main goal of this procedure was made to assess if the NPs were fixed onto the fabrics, even after a continuous contact with water and under agitation.

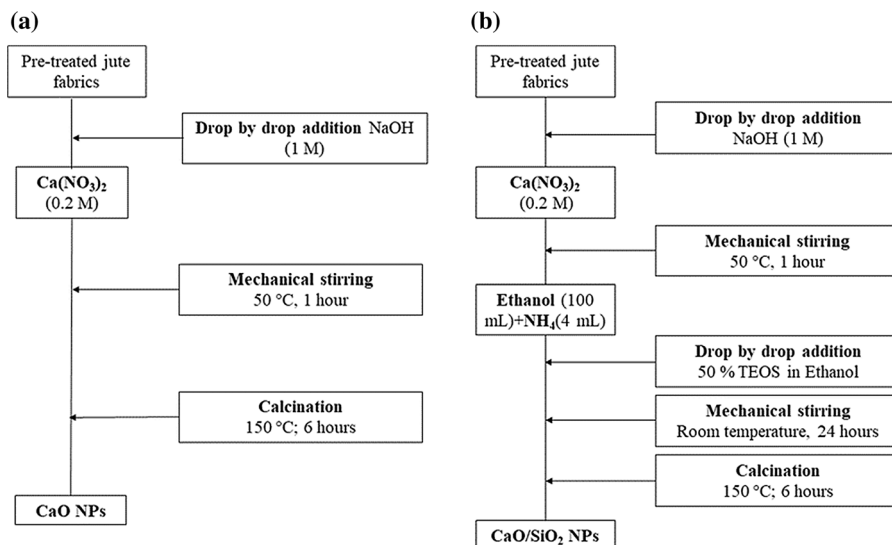
To understand if the CaO and the CaO–SiO<sub>2</sub> NPs were still present in the jute surface after the washing tests, the colour change ( $\Delta E$ ), using the jute fabric as standard was determined before and after the fabrics washing process. The  $\Delta E$  was assessed in distinct places of the samples (three replicates) with a Datacolour spectrophotometer using the Cielab coordinates D65/10 software (Lucerne, Switzerland).

## **Results and discussion**

In situ synthesis of the metal oxide nanoparticles onto the jute fabrics and respective characterization

As referred before, pre-treated jute fabrics were functionalized with CaO and SiO<sub>2</sub> NPs by in situ synthesis. The experimental procedure is summarized in Fig. 1.

It should be remarked that common synthesis procedures of CaO NPs include the use of higher reducing agents’ concentrations, higher calcination temperatures ( $> 500$  °C) as well as longer reaction times. In this work, as Fig. 1 shows, the synthesis protocols were optimized in order to obtain these metal oxide NPs using the simplest approaches and also at the same time (one pot synthesis) reducing reactions times and reagents. As the following sections will show, the adopted experimental procedure was successful for the obtention of jute fabrics functionalized with hierarchical structures of CaO–SiO<sub>2</sub> NPs.

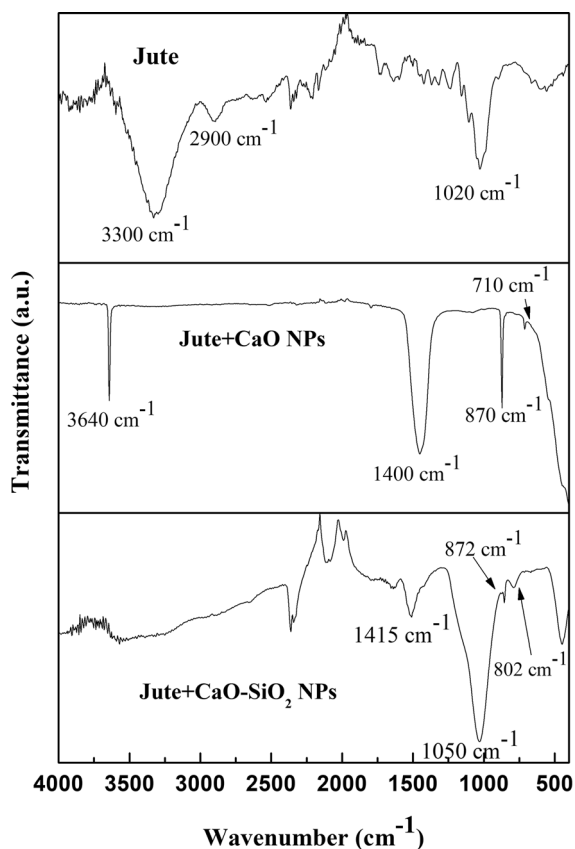


**Fig. 1** Schematic diagram of the experimental procedure for **a** in situ synthesis of CaO NPs onto the fabrics and **b** in situ synthesis of CaO-SiO<sub>2</sub> onto the fabrics

#### Attenuated Total Reflectance-Fourier Transform Infrared Spectroscopy (ATR-FTIR)

Figure 2 exhibits the ATR-FTIR spectra of jute fabric, jute fabric functionalized with CaO NPs and jute with CaO-SiO<sub>2</sub> NPs.

The jute fabric spectrum shows the presence of the typical bands of natural cellulosic fibers: cellulose, hemicellulose and lignin. It's possible to see the appearance of a band, at approximately 3300 cm<sup>-1</sup>, corresponding to the O-H stretching, related to the hydroxyl groups from cellulose and lignin and also to the existence of adsorbed water molecules (P Ferreira et al. 2018; Zouheyr et al. 2015). The band peaking at ~ 2900 cm<sup>-1</sup>, can be attributed to the asymmetric C-H stretching vibration of cellulose and hemicellulose (P Ferreira et al. 2018; Zouheyr et al. 2015) and the one peaking around 1020 cm<sup>-1</sup> is related to the C-O stretch vibration (Silva et al. 2016). The in situ synthesis of CaO NPs onto the jute, resulted in the appearance of the characteristic fingerprint peaks of CaO metal oxide, namely: O-H from Ca(OH)<sub>2</sub>, C-O, Ca-O-Ca and Ca-O bondings at 3640 cm<sup>-1</sup>, 1400 cm<sup>-1</sup>, 870 cm<sup>-1</sup> and 710 cm<sup>-1</sup> respectively (Anantharaman et al. 2016; Bharathiraja et al. 2018; Galván-Ruiz et al. 2009; Tangboriboon et al. 2012). In addition, due to the high reflective index of CaO NPs coating (S Anesh et al. 2015) onto the jute surface, there is no evidence of the peaks related with the

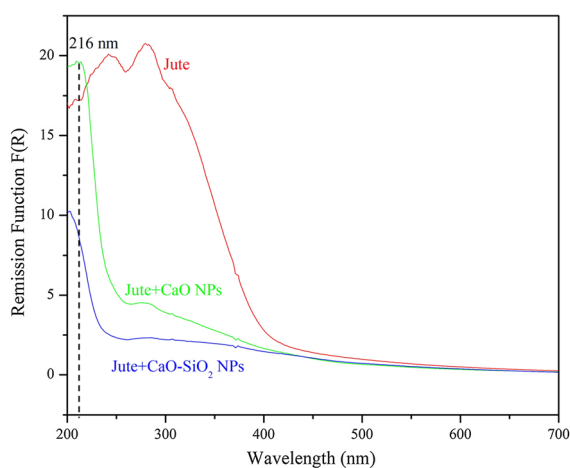


**Fig. 2** ATR-FTIR spectra of jute fabric, jute functionalized with CaO NPs and jute with CaO-SiO<sub>2</sub> NPs

natural fiber, possibly indicating the full covering of the analysed surface with the NPs, which is not allowing the appearance of cellulose characteristic bands observed in the jute spectrum. The SiO<sub>2</sub> NPs addition, led to the appearance of two new bands peaking at 1050 cm<sup>-1</sup> and 802 cm<sup>-1</sup> correspondent to asymmetrical stretching vibration of the Si–O–Si bonds, characteristic of the SiO<sub>2</sub> NPs formation (Li et al. 2014; El-Naggar et al. 2017). Due to the high intensity of the 1050 cm<sup>-1</sup> band, the fingerprint bands of CaO (at 1415 and 872 cm<sup>-1</sup>) are present in the spectrum, however with lower intensity when compared with the spectrum of jute + CaO NPs. In the spectrum of jute + CaO–SiO<sub>2</sub> NPs is also visible a very broad band between the 3200 and 3500 cm<sup>-1</sup>. The absorption bands peaking in this wavenumber range correspond to the stretching vibration of the O–H or Si–OH bonds, possibly indicating the establishment of hydrogen bonds between the hydroxyl groups of the jute and the NPs (Ferreira et al. 2016; El-Naggar et al. 2017). The existence of these hydrogen bonds leads to the conclusion that the NPs are anchored to the jute fabric surface.

#### Ground-State Diffuse Reflectance (GSDR)

The Kubelka–Munk remission functions of the jute fabrics without any functionalization and functionalized with CaO and with CaO–SiO<sub>2</sub> NPs are shown in Fig. 3.



**Fig. 3** GSDR spectra of the non-functionalized jute fabric (red), jute fabric functionalized with CaO NPs (green) and jute fabric functionalized with CaO–SiO<sub>2</sub> NPs (blue)

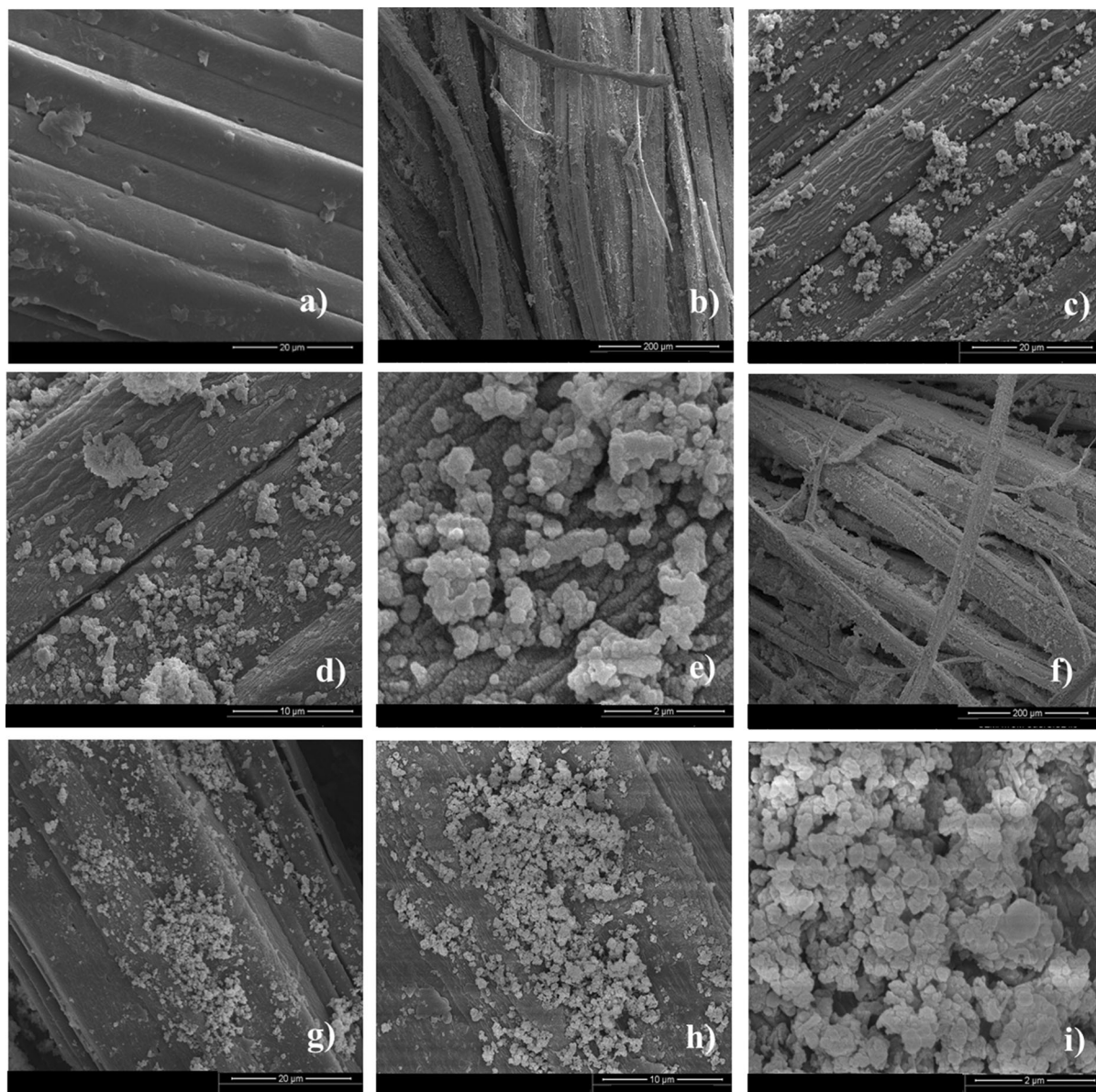
The fabric functionalized with CaO NPs present a new peak, when compared with the spectra of the non-functionalized fabric, at approximately 216 nm. This peak is probably related to the presence of the CaO NPs, confirming once again the successful incorporation of these NPs onto the fabric's surface. It's also possible to observe that, for the samples with NPs in their surface, there's a decrease in the absorption band intensity, when compared with the jute fabric without NP's. There's a decrease in the absorption intensity related to an increase of the reflectance. This phenomenon can be attributed to the presence of the NPs onto the surface of the fabrics, since they are inorganic white-pigments at nanoscale, which intrinsically present a high reflective index, easily reflecting the UV rays (S Anesh et al. 2015). The intensity of band related to the CaO NPs also decreases for the samples with CaO–SiO<sub>2</sub> NPs, this can be explained by the loss of some CaO NPs in the process of the addition of the SiO<sub>2</sub> NPs (Costa et al. 2018).

#### Field Emission Scanning Electron Microscopy (FESEM) and Energy Dispersive Spectroscopy (EDS)

The surface morphology of the pre-treated jute fabric and the jute fabric functionalized with CaO and CaO–SiO<sub>2</sub> NPs by in situ synthesis were evaluated by FESEM. EDS analysis of these samples was also performed in order to evaluate their elemental composition and the presence of Silica (Si) and Calcium (Ca) at the fabrics surface (Fig. 4 and Fig. 5).

Figure 4a exhibits the smooth and cleaned pre-treated jute fabric surface without NPs for comparison purposes. Figure 4b–e show the distribution of CaO NPs onto the fabric surface using different magnifications (from 200 to 2 μm). As the images show, the NPs have a spherical shape distributed all over the fiber surface with the presence of some aggregates. From FESEM images in Fig. 4f–i, it might be observed the presence of both NPs onto the fibers surface. In fact, the particles morphology is the same (spherical shape) and they are also covering all the surface. The obtained EDS spectra of jute, jute + CaO NPs and jute + CaO–SiO<sub>2</sub> NPs is presented in Fig. 5.

As expected, the EDS analysis confirms the existence of the peaks related with the presence of Ca and Si onto the jute fabrics surface. Through the ATR-FTIR, GSDR and FESEM-EDS analysis is possible to confirm that the NPs were synthesized successfully



**Fig. 4** FESEM images of **a** jute fabric surface; **b–e** images of jute fabric functionalized with CaO NPs using different magnifications respectively (200, 20, 10 and 2  $\mu\text{m}$ ) and **f–i** jute fabric functionalized with CaO–SiO<sub>2</sub> NPs (200, 20, 10 and 2  $\mu\text{m}$ )

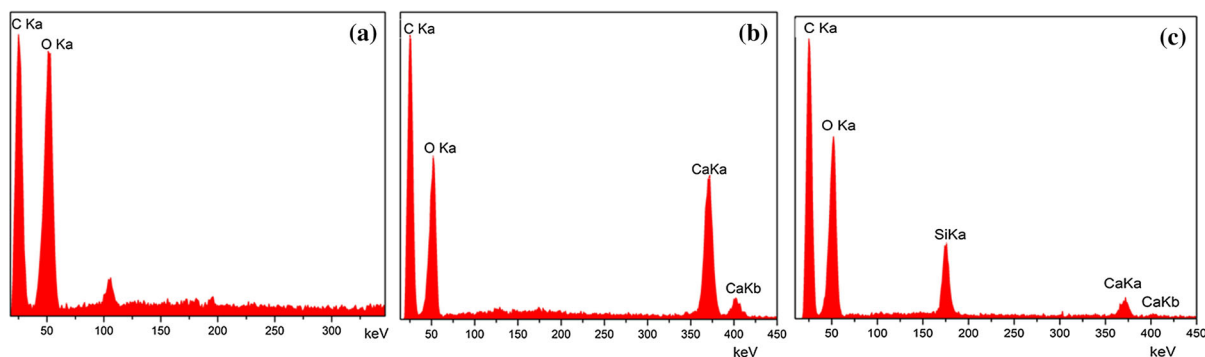
onto the surface of the jute fabrics. Most syntheses reported in literature use very high calcination temperatures, between 500 and 700 °C (Alavi and Morsali 2010; Anantharaman et al. 2016; Bharathiraja et al. 2018). In this work, the calcination temperature used was 150 °C, a lot less than the ones already reported. With this work, it was demonstrated that this temperature is enough to effectively obtain CaO NPs.

The use of lower temperatures makes this synthesis more sustainable and easier to apply at industrial scale.

#### *Thermogravimetric analysis (TGA)*

The previous techniques allowed the investigation of the morphology and qualitative chemical information of the synthesized NPs. Thermogravimetric analysis (TGA) can be used for the quantitative analysis and





**Fig. 5** EDS analysis of **a** jute fabric surface; **b** jute + CaO NPs and **c** jute + CaO–SiO<sub>2</sub> NPs

give us information about the amount of the NPs onto the fabric surface.

For this purpose, the jute fabric sample and the sample of jute with CaO–SiO<sub>2</sub> NPs were tested. The samples were subjected to a heating process from 25 to 600 °C and all the weight losses were considered giving special attention to the percentage of residues of each sample at the end of the heating process. The percentage of CaO–SiO<sub>2</sub> NPs was calculated using the equation system referred before. The jute fabric without NPs presented 1.62 wt% of residues, and the jute fabric functionalized with CaO–SiO<sub>2</sub> NPs presented 2.68 wt%. Considering that the final sample is composed by the jute fabric and the NPs, the percentage of NPs present at the fabric surface is approximately 1.06 wt%.

#### Functional properties evaluation

##### *Hydrophobicity properties*

The hydrophobicity of the jute fabric's surface was evaluated by the measurement of the WCA in 10 different places of the samples, in order to ensure homogeneity. If the obtained WCA is smaller than 90°, the surface is considered hydrophilic, but if the WCA is higher than 90°, the surface is considered hydrophobic. For a surface to be considered super-hydrophobic, the WCA has to be bigger than 150° (Marmur et al. 2017). The results of these measurements are shown in Fig. 6.

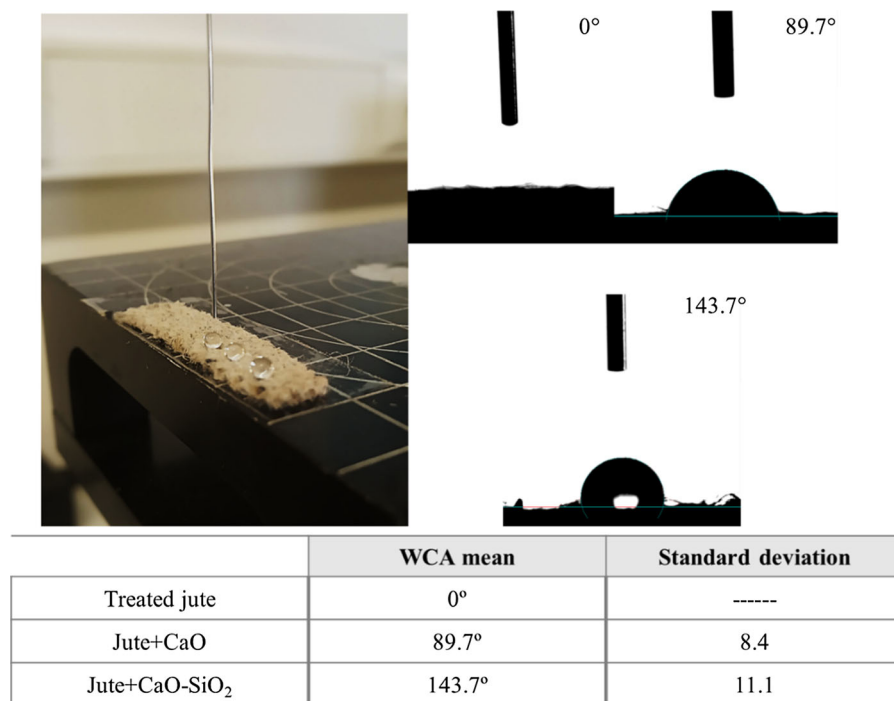
Jute fabric is hydrophilic by nature with a WCA of 0°. The incorporation of CaO NPs onto the jute fabrics' surface leads to a WCA of 89.7°, so it's possible to say that we started with a hydrophilic

surface, and with the functionalization of the fabric with CaO NPs we obtained a hydrophobic surface. However, with the addition of the SiO<sub>2</sub> NPs to the jute + CaO NPs systems, there's an increase in the WCA, reaching almost super-hydrophobicity (143.7°). This result is in accordance with the well know hydrophobic character of SiO<sub>2</sub> NPs (Jhatial et al. 2019; Seyfi et al. 2016) and can be very useful for self and easy cleaning applications.

##### *UV protection*

The UV protection properties of the developed systems were also evaluated. In order to do this, the UPF, UVA and UVB blocking were calculated, recurring to the standard EN13758-1. These values, for the non-functionalized pre-treated jute fabric and for the jute fabric functionalized with CaO and CaO–SiO<sub>2</sub> NPs are presented in Table 1.

From the results of Table 1, it's clear that the jute fabric without any kind of NPs have almost no UV protection, presenting a UPF of 8. Nevertheless, metal oxide NPs are used as UV blocking agents (Nalathambi 2011), and the obtained results for the fabrics functionalized with the NPs are in conformity with this information. The UV protection capability of metal oxides can be explained by several approaches. One of them states that this type of NPs reflects and/or scatters the UV rays, and another one focus one the semiconductive properties of metal oxides, leading to UV radiation absorption. This mechanism for semiconducting inorganic materials can be described by the use of photon energy in order to excite electrons from the valence to the conduction band (Bashari et al. 2019).



**Fig. 6** WCA measurements of the non-functionalized pretreated jute fabric, jute fabric functionalized with CaO NPs and jute fabric functionalized with CaO–SiO<sub>2</sub> NPs. Representative image of the jute fabric functionalized with CaO NPs

**Table 1** UPF, classification and UVA and UVB blocking (%) values of the jute, jute + CaO and jute + CaO–SiO<sub>2</sub> NPs samples

Sample	UPF	Classification	UVA blocking (%)	UVB blocking (%)
Jute	8	Low protection	82.1	89.6
Jute + CaO NPs	159	Excellent	99.3	99.4
Jute + CaO–SiO <sub>2</sub> NPs	425	Excellent	99.8	99.8

With the functionalization of the fabrics with the CaO NPs, the UPF value increases to 159, which means that these are classified as excellent, for UV protection purposes. For the jute fabric functionalized not only with CaO but also with SiO<sub>2</sub> NPs, this UPF value increased even more, reaching 425. Thus, the addition of SiO<sub>2</sub> NPs to the functionalization of the fabrics, improves the UV protection properties of the textiles. This result, once again, confirms the enhancement of the CaO NPs properties, with the addition of the SiO<sub>2</sub> NPs (Peerzada et al. 2019).

#### Antibacterial activity

Several studies have reported the use of plant extracts as effective antimicrobial agents, namely, extracts of the leaves and aerial parts of *Corchorous olitorius* (origin of jute fibers) (Sümmengen Özdenefe et al. 2019). At the same time, the antimicrobial effect of bast fibers (like hemp, bamboo and flax) has already been reported (Cai and Ma 2013). Jute fibers present a lignin content of approximately 16%, which is a very high value when compared with other bast fibers, such as hemp or flax, with a lignin content less than 5% (del Río et al. 2009). In this specific case, the raw jute fabrics that were used, were only treated with detergent, in order to eliminate the dirt and waxes present

on the fiber's surface. Thus, the chemical composition of the fibers was not changed (contrary to what happens in the alkali treatment, for example), having still a great quantity of lignin. Some studies already demonstrated that lignin has an antimicrobial effect, since the phenolic hydroxyl and methoxy groups that constitute it are biologically active (Alzagameem et al. 2019). This can explain the potential antibacterial activity of raw jute fibres.

Accordingly, the antibacterial activity of jute fabrics and jute fabrics functionalized with CaO and CaO–SiO<sub>2</sub> NPs was assessed in accordance with Japanese Standard JIS L 1902:2002. Gram-negative (*Escherichia coli*) and Gram-positive (*Staphylococcus aureus*) bacteria were tested and the results are presented in Table 2.

Typically, an antimicrobial agent may possess either bactericidal or bacteriostatic properties. Bactericidal activity implies the forthright killing of the organisms (killing of > 99.9%) while bacteriostatic activity is defined as maintenance of the original inoculum concentration, which means that it prevents the multiplication of bacteria without destroying it. In this study, the three samples presented bacteriostatic activity against both microorganisms, but the enhancement of the antimicrobial properties is obtained with the addition of the CaO and SiO<sub>2</sub> NPs. The antibacterial activity of lignocellulosic fibers, with similar lignin contents to the jute fibers, has already been reported, where it was also demonstrated that the antibacterial activity is related to the phenolic content in the material (Guna and Ilangovan 2018; Ilangovan et al. 2020). This is in accordance with the obtained results, since the jute fibers utilized were in their raw form, therefore presenting a great content of lignin.

In fact, although the jute fabrics *per se* already have antimicrobial properties, when functionalized with CaO NPs present bactericidal properties against *S. aureus*. This is in accordance with the results of Roy and co-authors (Roy et al. 2013) when studied the

antimicrobial activities of CaO NPs against *Staphylococcus epidermidis* (Gram-positive) and *Pseudomonas aeruginosa* (Gram-negative). They observed that the bactericidal concentration of CaO NPs was quite lower for *Staphylococcus epidermidis* in comparison to Gram negative *Pseudomonas aeruginosa*. This can be due to the higher amount of negatively charged peptidoglycans that would make Gram positive bacteria more susceptible to such positively charged antimicrobials (Ren et al. 2009). This is also in accordance with the obtained results, as the antibacterial activity of the functionalized samples was more pronounced against the gram-positive bacteria, when compared with the gram-negative one. Although, the results of antibacterial activity against *E. coli* were also very good, once again demonstrating the great antimicrobial potential of these systems. It's also possible to conclude, that for this specific property, the addition of the SiO<sub>2</sub> NPs didn't had a significant impact, since the obtained antibacterial activity values, for the two types of bacteria, of the jute fabric with CaO is not very different that the ones obtained for the jute fabric with CaO and SiO<sub>2</sub> NPs. These results confirm that jute fibers, and in particular jute fabrics functionalized with CaO and CaO–SiO<sub>2</sub> NPs present great antibacterial properties, hence they can be used for multiple antibacterial applications, namely in personal protective equipment.

#### Methylene blue degradation tests

The photocatalytic activity of the CaO and CaO–SiO<sub>2</sub> NPs was evaluated through the monitorization by UV-spectroscopy of MB degradation. For this purpose, the functionalized jute fabrics (with CaO or CaO–SiO<sub>2</sub> NPs) were immersed onto the MB solution and kept in the dark for 10 min to ensure the adsorption–desorption equilibrium. After this time, the samples were irradiated by UV light and several aliquots of the MB

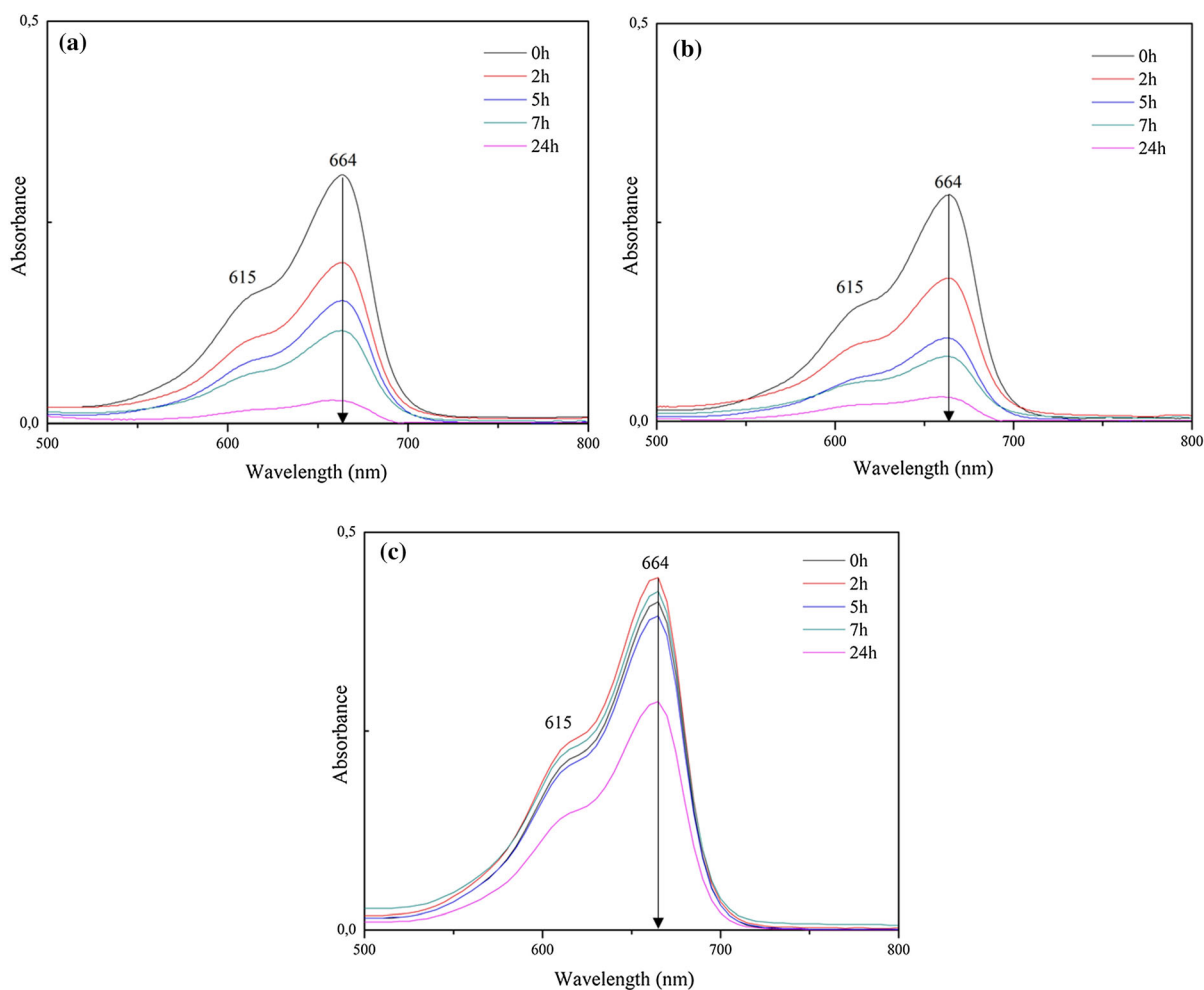
**Table 2** Antibacterial activity (%) values of the jute, jute + CaO and jute + CaO–SiO<sub>2</sub> NPs samples

Sample	<i>Staphylococcus aureus</i>	<i>Escherichia coli</i>
Jute	81.74 ± 9.50	95.84 ± 3.68
Jute + CaO NPs	99.98 ± 0.03	98.25 ± 2.38
Jute + CaO–SiO <sub>2</sub> NPs	99.96 ± 0.07	99.80 ± 0.29

solution (in contact with the fabric samples) were collected, after 2, 5, 7 and 24 h of UV radiation. For comparison purposes, the control solution of only MB in water was also monitored. Figure 7 show the obtained UV–Vis spectra of the three different samples under study: jute + CaO, jute + CaO–SiO<sub>2</sub> and control solution.

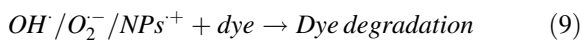
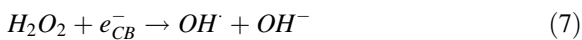
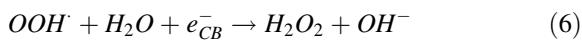
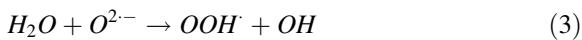
Through the study of the spectra of the two different types of NPs, it's visible that over time and with UV light exposure, the intensity of the band peaking at 664 nm (characteristic absorption band of MB (Yang et al. 2017)) decreased. When reaching the 24 h' time, the band related to the dye almost disappears, which indicates that the MB was degraded. When comparing the jute+CaO spectra with jute+CaO–SiO<sub>2</sub>, it may be

concluded that for the last system, the MB degradation is faster. For the 2-hour time, the absorbance at 664 nm for the jute+CaO was 1.3, yet, for the jute+CaO–SiO<sub>2</sub> was 0.9. Regarding the 7-hour time, the measured absorbance was 0.5 for jute+CaO and 0.2 for the jute+CaO–SiO<sub>2</sub>. This is probably related to the improvement of the photocatalytic activity of CaO, with the addition of the SiO<sub>2</sub> NPs, which is an accordance with literature, since SiO<sub>2</sub> improved photocatalysis has been reported for other metal oxide NPs (Guan 2005). It's also possible to observe in the control dye solution spectrum, that the intensity of the band peaking at 664 nm stays almost the same over time, which means that without the presence of the NPs the MB degradation doesn't occur.



**Fig. 7** Absorbance spectra of MB solution after 2, 5, 7 and 24 h of UV irradiation for **a** jute fabrics functionalized with CaO NPs, **b** functionalized with CaO–SiO<sub>2</sub> NPs and **c** for the MB solution (control)

The photodegradation of reactive dyes, using CaO and SiO<sub>2</sub> NPs as the photocatalyst has already been studied (Badr et al. 2008; Kornprobst and Plank 2012; Veeran et al. 2014). The degradation of specifically MB by this kind of NPs was also already demonstrated (Ameta et al. 2014). Thus, the mechanism of photocatalytic degradation of the CaO–SiO<sub>2</sub> NPs can be predicted:



Under UV radiation, the CaO–SiO<sub>2</sub> NPs molecules get excited, transferring one electron to the conduction band (1). This electron in the NPs' conduction band can reduce the molecular oxygen, producing a super oxide radical (2). The molecular oxygen adsorbed on the surface of the NPs, prevents the hole–electron pair recombination. This process decreases the rate of the photocatalytic degradation. In the presence of oxygen and organic molecules, the super oxide radical may form hydrogen peroxide or organic peroxide (3–5). Another path can also generate hydrogen peroxide (6) and this compound can form hydroxyl radicals, which are powerful oxidizing agents (7, 8). The produced radicals by this reaction are able of attack dye molecules and degrade them (9) (Ameta et al. 2014).

The degradation of MB was used as model reaction in order to evaluate the photocatalyst activity of the NPs, due to the obvious color change during the reaction (Yang et al. 2017). In future works, this photocatalytic activity is going to be evaluated for other dyes and toxic chemicals.

#### Durability test: wash fastness

For the washing test, the fabric samples were immersed in water and subjected to centrifugation treatment (to mimic the washing machine process as referred in the experimental part). The samples were dried and Cielab color coordinate, namely  $\Delta E$  was measured in three different sites of each sample (jute + CaO and jute + CaO–SiO<sub>2</sub>) and using the jute fabric as standard.

In Table 3, the average values of  $\Delta E$  before and after the washing test are presented.

For the jute sample loaded with CaO NPs, visible quantity of NPs was released into the solution resulting in a decrease of the  $\Delta E$  from 4.5 to 2.6 (color fading). This release is related with the weak attachment of the excess CaO NPs onto the jute fabric surface. However, after six hours of continuous washing there is still half of the NPs initial quantity at the jute surface. Regarding the sample of jute + CaO–SiO<sub>2</sub> NPs the variation is very small, almost undetectable (from 5.7 to 5.4). In fact, The SiO<sub>2</sub> NPs improved the attachment of the NPs to the fabric and the  $\Delta E$  values remained constant. SiO<sub>2</sub> NPs could be covalently bounded to the jute surface by silanes through –O–Si–O– or –CH<sub>2</sub>–O–Si–O– groups (Roe and Zhang 2009) disabling the release of the NPs onto the water as happened when only CaO NPs were present onto the jute. Overall, despite the variation exhibited by the sample with CaO NPs, the CaO–SiO<sub>2</sub> NPs fabric displayed an excellent performance after the washing test being the most promising sample under study.

**Table 3**  $\Delta E$  values for the Jute + CaO NPs before and after wash and for Jute + CaO–SiO<sub>2</sub> NPs before and after wash samples compared with jute fabric

Sample	$\Delta E$
Jute + CaO NPs before wash	4.5
Jute + CaO NPs after wash	2.6
Jute + CaO–SiO <sub>2</sub> NPs before wash	5.7
Jute + CaO–SiO <sub>2</sub> NPs after wash	5.4

## Conclusion

This work presents, for the first time in literature, the in situ synthesis of CaO and SiO<sub>2</sub> NPs onto jute fabrics, in order to obtain a multifunctional material. The two types of NPs were successfully synthesized onto the surface of the pre-treated jute fabrics, through sustainable and simple processes and materials.

ATR-FTIR, GSDR and FESEM-EDS results confirmed the presence of not only the CaO, but also the SiO<sub>2</sub> NPs onto the fabrics surface. The ATR-FTIR results, particularly, besides confirming the presence of the NPs onto the jute fabrics also showed that they were well anchored to the fabrics, since there's the establishment of hydrogen bonds between the hydroxyl groups of the jute and the NPs. The FESEM images also revealed that the NPs presented a spherical shape and a homogeneous distribution all over the jute surface. The developed systems presented great properties, namely hydrophobicity, UV protection, antibacterial activity and MB degradation. Regarding the hydrophobicity, the pretreated jute fabrics presented a hydrophilic character, with a WCA of 0°, nonetheless, the functionalization of the fabrics with CaO led to hydrophobicity, presenting a WCA of 89.7°. However, the addition of SiO<sub>2</sub> NPs increased the WCA of the jute fabrics, reaching almost superhydrophobicity, with a WCA of 143.7°. The functionalized jute fabrics also presented very good UV protection values, with UPF values of 50 + , and with UVA and UVB blocking values of almost 100%. In this case, the addition of the SiO<sub>2</sub> improved once again the properties of the systems, since the UPF obtained for the jute CaO sample was 159, increasing to 425 for the jute CaO–SiO<sub>2</sub> sample. Regarding the antibacterial activity, all the samples, including the non-functionalized jute fabrics presented bacteriostatic activity against two types of bacteria, *Escherichia coli* and *Staphylococcus aureus*, but an enhancement of the antimicrobial properties was observed with the addition of the CaO and SiO<sub>2</sub> NPs, reaching values of 99.8% of antibacterial activity against the two microorganisms. The developed samples also showed photocatalytic activity, being able to degrade MB. Once again, the presence of the SiO<sub>2</sub> NPs improved the obtained results as the MB degradation was faster for the sample with these NPs, when compared with the samples with only CaO NPs). The wash durability of the nanocoating was also

evaluated, confirming that the NPs were well attached to the fabrics and that the SiO<sub>2</sub> NPs improved this anchorage.

Ultimately, this work shows the development of a multifunctional fibrous system based on jute and CaO and SiO<sub>2</sub> NPs, that can be used in several applications, having a specific potential for the application in personal protective equipment's, namely for protection against noxious chemical and biological agents, since the developed systems present the ability to degrade a toxic dye, which can possibly translate in the degradation of other harmful chemicals and present great antibacterial activity against gram-negative and gram-positive bacteria.

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**Authors' contributions** Joana C. Araújo was responsible for the experimental part and wrote the paper. Diana P. Ferreira supervised the work and also wrote the paper. Pilar Teixeira performed the antibacterial activity tests and participated in the writing of the paper. Raul Figueiro also supervised the work.

## Compliance with ethical standards

**Conflict of interest** The authors declare no conflict of interest.

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