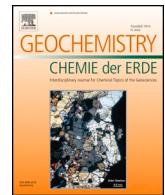




## Geochemistry

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# Geochemistry of granitic aplite-pegmatite dykes and sills and their minerals from the Gravanho-Gouveia area in Central Portugal

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## ARTICLE INFO

**Keywords:**

Variscan granites  
Pegmatites  
Feldspars  
Micas  
Fractionation series  
Cassiterite  
Hydrothermal fluids

## ABSTRACT

Two distinct series of Variscan granitic rocks have been distinguished in the Gravanho-Gouveia area of Portugal, based on field work, variation diagrams for major and trace elements, rare earth patterns and  $\delta^{18}\text{O}$  versus total FeO diagram of rocks, anorthite content of plagioclase, BaO and  $\text{P}_2\text{O}_5$  contents of feldspars and Al<sup>VI</sup> versus Fe<sup>2+</sup> diagram for magmatic muscovite. One series consists of a late-orogenic porphyritic biotite > muscovite granite (G1), less evolved beryl-columbite pegmatites and more evolved beryl-columbite pegmatites showing gradational contacts. The other series consists of post-orogenic porphyritic muscovite > biotite granodiorite to granite (G2), slightly porphyritic muscovite > biotite granite (G3) and lepidolite pegmatites. In each series, pegmatites are derived from the parent granite magma by fractional crystallization of quartz, plagioclase, K-feldspar, biotite and ilmenite. Some metasomatic effects occur like muscovite replacing feldspars, chlorite in pegmatites of the first series and a late muscovite in pegmatites of the second series, probably due to hydrothermal fluids. The lepidolite pegmatites contain cassiterite and two generations of rutile. The first magmatic generation consists of homogeneous crystals and the second generation occurs as heterogeneous zoned crystals derived from hydrothermal fluids. The beryl-columbite pegmatites and lepidolite pegmatites also contain the first magmatic generation and the late hydrothermal generation of zoned columbite-group minerals. More evolved beryl-columbite pegmatites were converted into episyenite by intense hydrothermal alteration and regional circulation of fluids in the granitic rocks.

## 1. Introduction

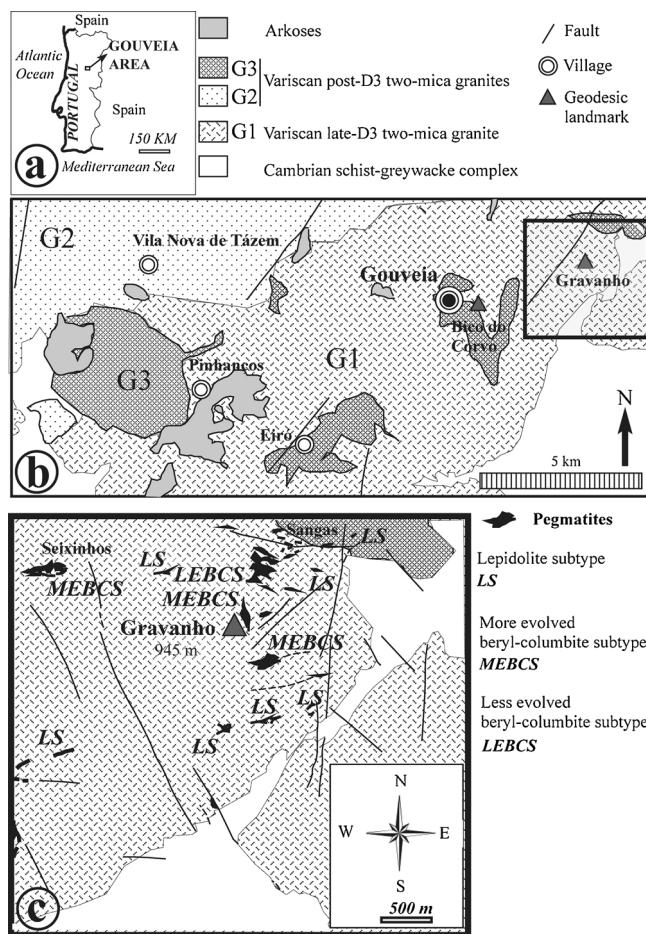
Granitic pegmatites, their inner-zonings and minerals have been studied by several authors (e.g., Černý et al., 1985a, b, 2005; London, 2008; London and Morgan, 2012; Neiva, 2013; Neiva and Ramos, 2010; Roda-Robles et al., 2012, 2016). The origin of granitic pegmatites is mainly attributed to the differentiation of a granite magma. However, a few other authors have contested that origin for LCT pegmatites (Černý and Ercit, 2005) and suggest that they result from anatexis (e.g., Stewart, 1978; Bongiolo et al., 2016; Simons et al., 2016).

Granitic aplite-pegmatites and their minerals from three areas of central Portugal, in the Central Iberian Zone, have been studied and result from fractional crystallization of quartz, plagioclase, K-feldspar, biotite and ilmenite of a two-mica granite magma in each area, supported by modelling of major and trace elements (Neiva et al., 2008, 2012; Neiva and Ramos, 2010).

The fourth area chosen for study, the Gravanho-Gouveia area, is significantly different from those already studied, the Arcozelo da Serra area, Guarda-Belmonte area and Sabugal area, because pegmatites often are metasomatically altered and episyenitization occurs in the area. In general, it is difficult to obtain chemical analyses of metaso-

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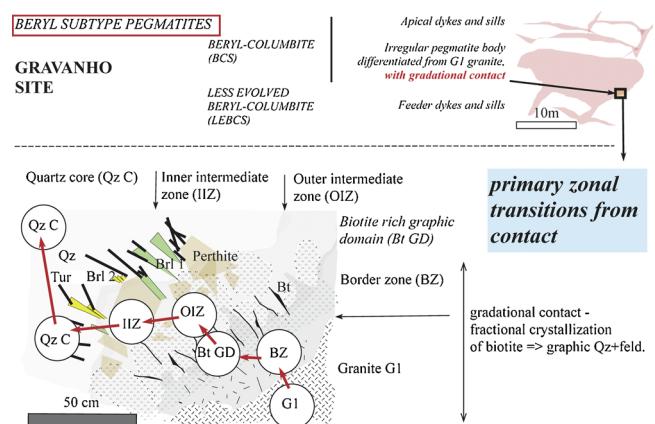
E-mail address: [neiva@dct.uc.pt](mailto:neiva@dct.uc.pt) (A.M.R. Neiva).



**Fig. 1.** Geological framework and field relations for granites and pegmatites. a. location of the studied area; b. field relations between granites and host-rocks enhancing the location of the major pegmatite field at the surroundings of Gravanho landmark; c. field distribution of the outcropping main types of pegmatites.

matic pegmatites and determine from which primary minerals the secondary minerals were derived. Therefore, papers on this subject are rare, as mentioned in London (2008).

The present paper reports the mineralogy, geochemistry and petrology of granites, dykes and sills composed of both aplite and pegmatite, which are referred to as aplite-pegmatites or simply pegmatites, from the Gravanho-Gouveia area to understand their origin. Whole-rock hydrothermal alteration of pegmatites is also studied.



**Fig. 2.** Sketch of the vertical section for the structure of *in situ* differentiated inner-granite, beryl subtype pegmatites, in the Gravanho site, with detail of gradational transition between the granite G1 and the zoned pegmatite. Qz – quartz, Brl – beryl (1- blue, 2- yellow); Tur – tourmaline; Bt – biotite. BZ – Border Zone, BGD – Biotite-rich Graphic Domain, OIZ – Outer Intermediate Zone, IIZ – Inner Intermediate Zone, Qz C – Quartz Core. Mineral abbreviations are as in Whitney and Evans (2010).

## 2. Geology

The Iberian Massif corresponds to the south-western extension of the European Variscan Belt. This massif is rich in Carboniferous and Permian plutonic rocks (e.g. Castro et al., 2002). The Gravanho-Gouveia area is part of the Central Iberian Zone, which is the innermost zone of the Iberian Massif (e.g. Farias et al., 1987).

In the Gravanho-Gouveia area, granites intruded the Cambrian schist-greywacke complex. The age of these granites was determined by the U-Th-Pb monazite method using SHRIMP (Neiva et al., 2009). The age of the coarse- to very coarse-grained late-orogenic porphyritic biotite > muscovite granite (G1) is  $301.4 \pm 2.6$  Ma, while the coarse-grained porphyritic muscovite > biotite granodiorite to granite (G2) and medium- to coarse-grained slightly porphyritic muscovite > biotite granite (G3) are post-orogenic (Fig. 1a, b). The age of the granite G2 is  $288.6 \pm 2.5$  Ma. The contacts between granite G1 and granites G2 and G3 are sharp, while granite G2 passes gradually to granite G3.

Two principal types of aplite-pegmatite dykes and sills intruded the granite G1: (i) aplite-pegmatites of the beryl-columbite subtype (rare-element class) that form larger irregular bodies, and (ii) lepidolite subtype pegmatites (LCT complex type, Černý and Ercit, 2005) that form numerous tabular bodies (Fig. 1c, Table 1). The relatively less evolved beryl-columbite aplite-pegmatites form dykes and sills that predominate at the lower topographic levels of the inner-granite pegmatite field (Fig. 2). Locally they are followed upwards to the confluence with irregular more evolved beryl-columbite aplite-pegmatites,

**Table 1**

Pegmatite characteristics at the outcrop scale from de Gravanho-Gouveia area, central Portugal.

aplite- pegmatite subtypes	approximate number in pegmatite field	shape	size - vertical width / extension	internal structure
LS pegmatites	28	tabular	< 2 m / 10 – 100m	layered
MEBCS pegmatites	2	irregular	5-20 m/40 m	zoned
LEBCS pegmatites	11	tabular to lens-shaped	1-5 m / 30 – 150m	homogeneous with comb-border

LEBCS – less evolved beryl-columbite; MEBCS – more evolved beryl-columbite; LS – lepidolite.

**Table 2**

Mineral assemblages in differentiated internal structures of aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

aplite- pegmatite subtypes	internal units	essential minerals (abundance: + to + + + + +)	crystal sizes	approximate sequence of crystallization
Lepidolite	border comb structure	muscovite (+ + +)	< 3cm	muscovite ± microcline = > albite ± quartz
		perthite (+ +)	< 3cm	
		albite (++)	< 3cm	
		quartz (++)	< 2cm	
	inner layers	albite (+ + to + + + +)	< 2cm	topaz ± montebrasite = > lepidolite ± albite = > lepidolite ± quartz
		lepidolite (< + + + +)	< 1cm	
		topaz( + )	< 6cm	
		montebrasite-ambligonite (+)	< 6cm	
		quartz (++)	< 6cm	
		biotite (++)	< 3cm	
More evolved beryl-columbite	border transition to outer intermediate zone	perthite /quartz graphic(+ + + +)	< 60 cm	biotite = > muscovite = > graphic assemblage = > perthite ± albite ± quartz
		biotite ( + )	< 30cm	
		perthite /quartz graphic(+ +)	< 40cm	
		muscovite (++)	< 15cm	
		perthite (++)	< 40cm	
		albite (++)	< 3cm	
		quartz ( + )	< 5cm	
		perthite ( + + )	< 35cm	
		muscovite ( + )	< 20cm	
		albite (++)	< 4cm	
Less evolved beryl-columbite	border inner granular to aplitic groundmass	quartz ( + + + +)	< 50cm	perthite ± albite ± muscovite = > quartz ± beryl = > tourmaline beryl = > tourmaline = > quartz biotite ± tourmaline = > muscovite = > albite ± perthite ± quartz tourmaline = > perthite = > albite = > quartz = > muscovite
		beryl( + )	< 15cm	
		tourmaline ( + )	< 5cm	
		tourmaline ( + + )	< 6cm	
		biotite ( + + )	< 6cm	
		muscovite ( + + + )	< 2cm	
		albite ± perthite ± quartz ( + + + + )	< 2cm	
		perthite ( + + + )	< 1cm	
		albite ( + + + )	< 0.5cm	
		quartz ( + + + )	< 0.5cm	

Note: crystal sizes are suggested by major lengths of their observed sections in outcrops.

which are differentiated *in situ*. The border structures of feeder dykes and sills indicate differential movement of pegmatite melt through the granite mass with contrasting rheological behaviour. At the feeder veins position, small-scale magmatic structures illustrate the contrasting rheology and motion of pegmatite magmas relative to the granite G1 viscous groundmass. Sharp contacts and host granite feldspar phenocrysts and biotite lamellae are parallel to the pegmatite elongation. This suggests a shear induced differential flux displacement of a pegmatite mush in relation to a incompletely consolidated granite host. At the upper positioned and irregular pegmatite bodies, the gradual transition between granite and pegmatite indicates *in situ* crystallization and a much more stationary and dynamically stable granite-pegmatite boundary. In contrast, a more stationary inner fractionation in irregular pegmatite bodies produces a centripetal and gradually enlarged inner growth of prismatic and elongated comb crystals, and a centrifugal and gradational transition to the normal grain size of the host granite. Feeders of less evolved beryl-columbite aplite-pegmatites are homogeneous, although an incipient comb structure of black tourmaline

might be present at the pegmatite inside the border and nucleated over the sharpest contacts. Zoned primary transitions, from contact inward, occur mainly at the more evolved beryl-columbite aplite-pegmatites and are interpreted as *in situ* crystallization of a residual melt, without significant compositional or thermodynamic gradients towards the host granite and without signs of relative displacement regarding the emplacement. Only the more evolved beryl-columbite aplite-pegmatites are clearly zoned (Fig. 2, Tables 1 and 2).

The lepidolite (LS) pegmatite sills crosscut irregular bodies of beryl-columbite pegmatites at their apical margins at the upper topographic levels of the pegmatite field. Therefore, they are younger than the less and more evolved beryl-columbite aplite-pegmatites (Fig. 1c). LS sills from the same generation in direct contact with the granite G1 show interface textures typical of non-equilibrium such as strips of zinnwaldite replacing biotite outwards from both contacts in the surrounding granite G1. LS aplite-pegmatites, intruded at the upper granite G1 levels, may reveal a comb layered inner structure of alternating predominantly albite and lepidolite layers. The content of quartz increases

towards the core, while muscovite and K-feldspar contents are significant only at the border, locally forming a distinct comb (Table 2). In these pegmatites, tourmaline occurs as rare brown crystals 1 mm in size, or as light coloured (locally gemmy, mainly pinkish in colour) crystals (several centimetres in length and less than a centimetre in width), in some innermost miarolitic voids in association with later cookeite and infilling clay minerals (Leal Gomes, 2010).

### 3. Analytical methods

The major and most trace elements of granites and aplite-pegmatite dykes and sills were determined using X-ray fluorescence (XRF) and the Tertian and Claisse (1982) method, with a precision better than  $\pm 1\%$  for major elements and Rb, and  $\pm 4\%$  for other minor and trace elements. Lithium was determined by atomic absorption spectroscopy with a precision of  $\pm 2\%$ . Zinc, Cu and Pb were analysed by ICP-AES with a precision of  $\pm 5\%$ . Detection limits were 2.5 ppm for Li; 3 ppm for Nb, Zr, Y, Sr and Rb; 4 ppm for Cu; 5 ppm for Ta and Th; 6 ppm for Zn, Sn and Pb; 7 ppm for Ba. These determinations were carried out at LNEG, S. Mamede de Infesta, Portugal.

The total FeO of the granitic rocks was determined by titration with a standardized potassium permanganate solution with a precision of  $\pm 1\%$ . For this determination 0.5 g of a granitic sample was decomposed and dissolved with an acid mixture prepared in another crucible by adding 5 ml of concentrated H<sub>2</sub>SO<sub>4</sub> to 5 ml of ultra-pure water and then 5 ml of HF was added. The F content of these rocks was analysed by selective ion electrode with a precision of  $\pm 2\%$ . These determinations for granites were carried out at the Department of Earth Sciences, University of Coimbra, Portugal, whereas those for aplite-pegmatite dykes and sills were done at LNEG, Portugal. Loss on ignition (LOI) for rocks was determined following the method of Lechner and Desilets (1987). The REE concentrations were measured by ICP-MS with a precision of  $\pm 5\%$  in the SGS Laboratory, Toronto, Canada, for granites and at LNEG, for aplite-pegmatite dykes and sills and mutually consistent results by checking the same reference materials. Whole-rock oxygen isotope analyses were performed by gas mass spectrometry using the method of Clayton and Mayeda (1963), with the  $\delta^{18}\text{O}$  reproducibility of  $\pm 0.2\text{\textperthousand}$  using the quartz standard NBS-28, at the Department of Earth Sciences, the University of Western Ontario, Canada.

The major element compositions of minerals from granites were analysed using the Modified Cambrian Geoscan Link System energy-dispersive system, but selected feldspars and micas were analysed using a Cameca SX100 electron microprobe at the Department of Earth Sciences, University of Manchester, U.K. (Neiva et al., 2009). The electron microprobes were operated at an accelerating voltage of 15 kV and a 15 nA beam current. Detection limits (3 $\sigma$ ) above mean background were < 0.03 wt.% oxide for most components, except for F (0.1 wt.%), BaO (0.06 wt.%) and Cl (0.01 wt.%), with counting times of 80 s for F, Cl and BaO. The major element compositions of minerals from aplite-pegmatite dykes and sills were determined using a Cameca Camebax electron microprobe and also a Hyperprobe Jeol JXA-8500 F, operating at 15 kV accelerating voltage and 15 nA and 10 nA beam current, respectively, for the former and the latter instrument. Detection limits (3 $\sigma$ ) above mean background were 0.03 wt.% oxide, except for BaO, Rb<sub>2</sub>O and Cs<sub>2</sub>O (0.06 wt.%) and F (0.1 wt.%) with counting times of 80 s for BaO, Rb<sub>2</sub>O, Cs<sub>2</sub>O and F. After dissolution and dilution, Be and Li in beryl were determined by ICP-MS with a precision of  $\pm 2\%$  and H<sub>2</sub>O<sup>+</sup> was determined with a Penfield tube and a precision

of  $\pm 1\%$ . Chemical analyses of cassiterite and rutile were determined using a Cameca Camebax electron microprobe conducted at an accelerating voltage of 15 kV and a beam current of 20 nA. The signal of each element was counted for 20 s. Raw-data were corrected using the PAP procedure of Pouchou and Pichoir (1985). These determinations were carried out at LNEG, S. Mamede de Infesta, Portugal. The Li<sub>2</sub>O content of muscovite was calculated using the equation Li<sub>2</sub>O = 0.3935 F<sup>1.236</sup>, and those of zinnwaldite, trilithionite and poly lithionite using the equation Li<sub>2</sub>O = (0.289 x SiO<sub>2</sub>) - 9.658 (Tischendorf et al., 1997), but there are some drawbacks of this approach.

The general formula of Henry et al. (2011) for tourmaline was used. Structural formulae were calculated on the basis of 31 anions using the Microsoft Excel™ worksheet of Tindle et al. (2002). Boron was assumed to be stoichiometric (e.g., Hawthorne, 1996). All Fe and Mn were assumed to be divalent. The Li concentrations were calculated according to the multiple regression method of Pesquera et al. (2016). The obliquity  $\Delta = 12.5 (d_{131} - d_{131})$  was determined by X-ray diffraction patterns of separates of K-feldspar. The error does not exceed  $\pm 0.003$ .

### 4. Petrography

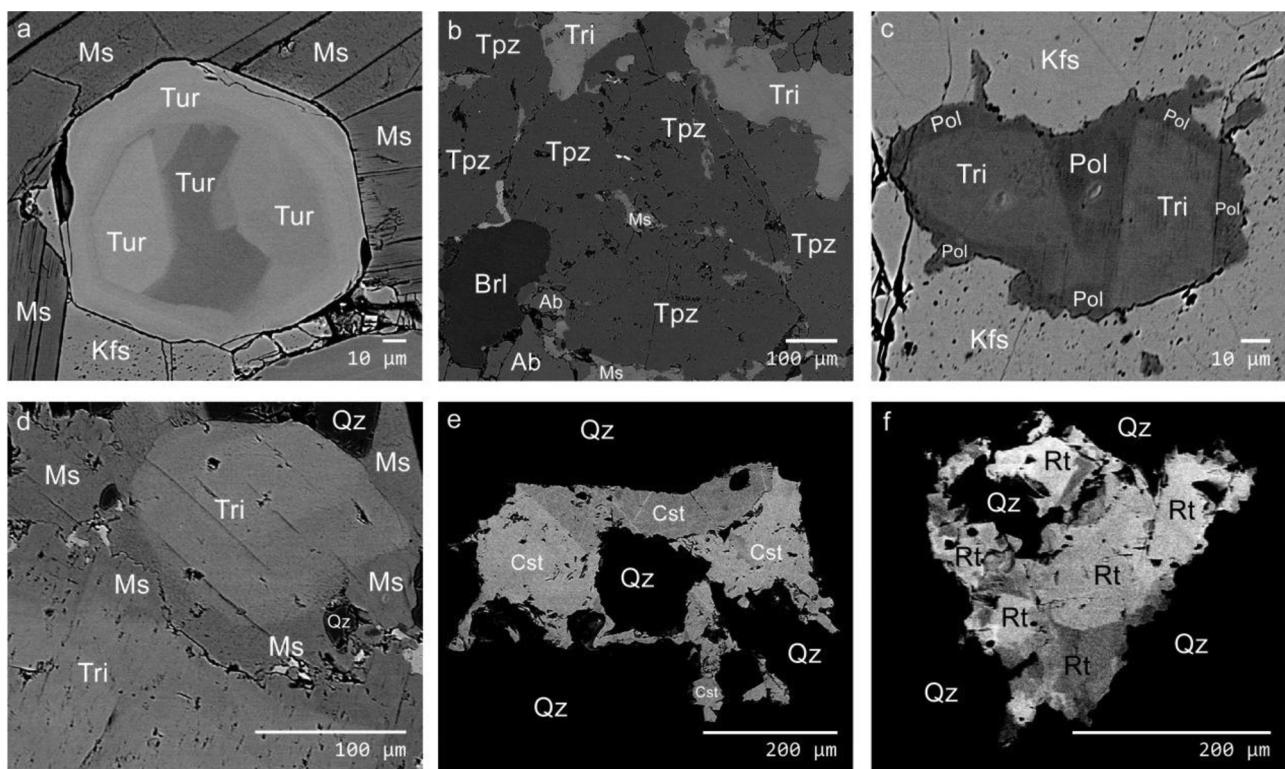
The granites and aplite-pegmatites contain the minerals indicated in Table 3. All granites have a subhedral granular texture and contain phenocrysts of K-feldspar and plagioclase. Anhedral quartz has undulose extinction, is fractured and contains inclusions of all minerals. K-feldspar and plagioclase are subhedral and show undulose extinction. Micas from the granite G1 show some orientation. Biotite and

**Table 3**

Minerals of granites and aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

	Granites			Aplite-pegmatites		
	G1	G2	G3	LEBCS	MEBCS	LS
Quartz	X	X	X	X	X	X
K-feldspar	X	X	X	X	X	X
Plagioclase	X	X	X	X	X	X
Biotite	X	X	X			
Muscovite	X	X	X	X	X	X
Zinnwaldite						X
Poly lithionite						X
Trilithionite						X
Late muscovite						X
Tourmaline				X	X	X
Andalusite	X	X	X			
Sillimanite	X	X	X			
Beryl					X	X
Topaz						X
Columbite-tantalite				X	X	X
Cassiterite						X
Rutile						X
Zircon	X	X	X	X	X	X
Monazite	X	X	X	X	X	X
Apatite	X	X	X	X	X	X
Ilmenite	X	X	X	X	X	X
Chlorite	X	X	X	X	X	X
Secondary muscovite	X	X	X	X	X	X

LEBCS - less evolved beryl-columbite, MEBCS – more evolved beryl-columbite, LS - lepidolite. Rare triple-zwieselite occurs in MEBCS and rare amblygonite-montebrasite occurs in LS. Both phosphates were identified by XRD. Poly lithionite and trilithionite correspond to the lepidolite series.



**Fig. 3.** Backscattered images of minerals from granitic aplite-pegmatites from the Gravanhó-Gouveia area, central Portugal. a. Zoned tourmaline crystal from a less evolved beryl-columbite pegmatite replaces muscovite and K-feldspar. b, c, d, e, f from lepidolite pegmatites. b. Topaz replaces muscovite; trilithionite penetrates topaz and beryl penetrates albite and topaz. c. Polylithionite penetrates and surrounds trilithionite; this crystal replaces K-feldspar. d. Zoned crystal with a core of trilithionite and a partial rim of a late muscovite. e. Zoned cassiterite grain. f. Heterogeneous rutile grain. Mineral abbreviations as in Whitney and Evans (2010): Tur - tourmaline, Ms - muscovite, Kfs - K feldspar, Tpz - topaz, Brl - beryl, Ab - albite, Tri - trilithionite, Pol - polylithionite, Qz - quartz, Cst - cassiterite, Rt - rutile.

muscovite are subhedral, intergrown and contain inclusions of andalusite, zircon, monazite, apatite and ilmenite. Needles of sillimanite occur in muscovite, they are rarer in feldspar and quartz and occur very rarely in andalusite from the granite G3. Biotite shows some chloritization. Some secondary muscovite replaces plagioclase.

In both the less and more evolved beryl-columbite aplite-pegmatites, quartz is anhedral with undulose extinction and is fractured. K-feldspar is subhedral and microperthitic microcline. Plagioclase is subhedral, surrounds K-feldspar and muscovite and replaces them. Muscovite is subhedral and contains inclusions of zircon, monazite and apatite. Tourmaline shows anhedral unzoned crystals and zoned crystals (Fig. 3a), which are pleochroic from greenish-brown to light brown. Tourmaline replaces K-feldspar, muscovite and quartz. The first magmatic generation and the second hydrothermal generation of columbite-tantalite were distinguished (Neiva et al., 2015). Some secondary muscovite replaces plagioclase. Subhedral chlorite occurs. In more evolved beryl-columbite aplite-pegmatites, crystals of beryl are subhedral, white, green, blue and yellow in colour. Ilmenite is subhedral and rare.

Lepidolite aplite-pegmatites contain anhedral quartz, showing undulose extinction and fractures. K-feldspar is subhedral and intensively microperthitic microcline. Plagioclase is subhedral, surrounds K-feldspar and muscovite and replaces them. Muscovite is subhedral and contains inclusions of zircon, monazite and apatite. Tourmaline is

anhedral, dark in colour and replaces quartz, K-feldspar and muscovite. Topaz contains inclusions of muscovite (Fig. 3b) and replaces muscovite. Trilithionite penetrates topaz (Fig. 3b) and contains inclusions of K-feldspar and albite; it locally replaces K-feldspar, and also is replaced by albite. Beryl is pink in colour and penetrates albite and topaz (Fig. 3b). Polylithionite surrounds and penetrates trilithionite (Fig. 3c). A late muscovite up to 1 cm in length surrounds trilithionite (Fig. 3d). A late generation of quartz replaces K-feldspar and trilithionite. Ilmenite is rare. Most cassiterite crystals are unzoned, but some show a faint zoning and several crystals show a darker zone and a lighter zone (Fig. 3e). Rutile occurs as homogeneous and heterogeneous grains. The latter have darker and lighter zones (Fig. 3f). Rutile occurs associated with zinnwaldite and apatite at the footwall of the dyke. The first magmatic generation and the second hydrothermal generation of columbite-tantalite were distinguished (Neiva et al., 2015). Some secondary muscovite replaces plagioclase.

Metasomatic rocks resembling episyenite occur in the area. Metasomatism of more evolved beryl-columbite aplite-pegmatites started with an increase in the obliquity of K-feldspar and potassic alteration around perthitic veinlets, and ended with an intensive microclinization and redness of these aplite-pegmatites, loss of quartz and volume and also occurrence of corroded zones filled by epidote, chlorite and hematite and locally by fluorite. Secondary muscovite also occurs.

## 5. Whole-rock geochemistry

Whole-rock samples were obtained by channel sampling, transversal to the strike of tabular, homogeneous to heterogeneous pegmatites. Representative chemical analyses and trace elements of granites and aplite-pegmatites from the Gravanho-Gouveia area are given in [Table 4](#).

The studied rocks form two distinct evolutionary trends: (i) from granite G1 through the less evolved to the more evolved beryl-columbite pegmatites, and (ii) from granite G2 through granite G3 to lepidolite pegmatites ([Fig. 4](#)). The less evolved beryl-columbite pegmatites and more evolved beryl-columbite pegmatites are distinguished mineralogically and chemically from the lepidolite pegmatites. The beryl-columbite pegmatites (analyses 4–7, [Table 4](#)) have a higher Rb content and lower TiO<sub>2</sub>, total FeO, MgO, CaO, Zr, Y, Sr, Ba and Th contents than the granite G1. Variation diagrams of these granitic rocks define the fractionation trends ([Fig. 4](#)).

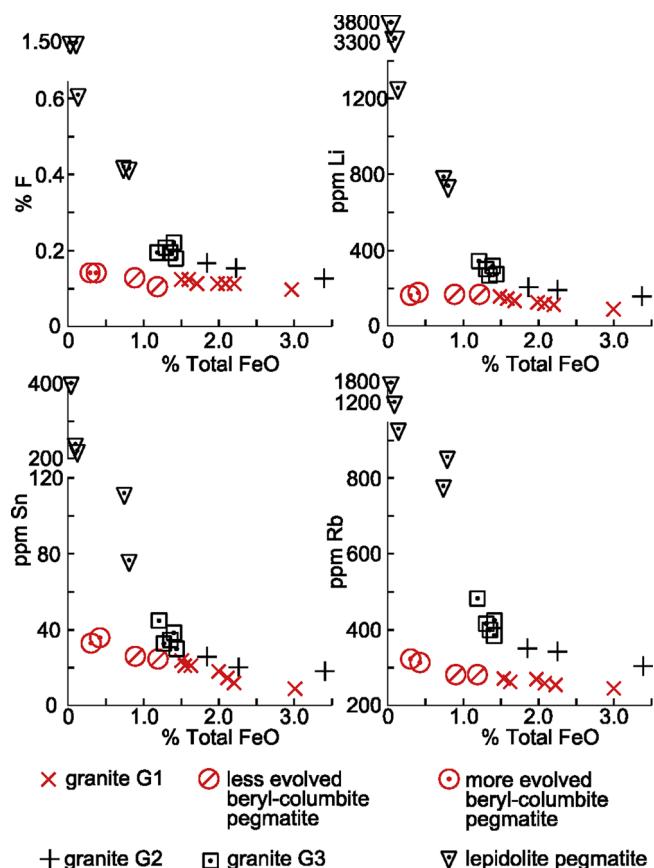
From the granite G2 to the granite G3 and to lepidolite pegmatites, the F, Sn, Li and Rb contents increase progressively while TiO<sub>2</sub>, total FeO, MgO, CaO, K<sub>2</sub>O, Zr, Y, Sr, Ba and Th contents decrease progressively (analyses 8–16, [Table 4](#)). These two granites and pegmatites define similar fractionation trends ([Fig. 4](#)). These pegmatites are the richest in F, Nb, Ta, Sn, Li and Rb in the studied area.

**Table 4**

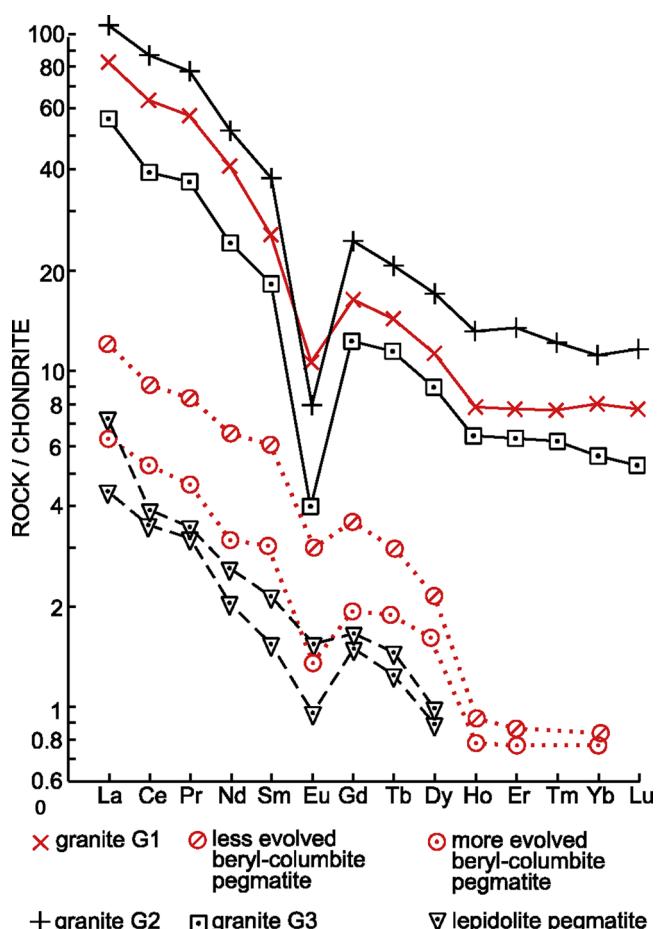
Representative chemical analyses in wt.% and trace elements in ppm of granites and some granitic aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

Granite G1			LEBCS pegmatites		MEBCS pegmatites		Granite G2		Granite G3		LS pegmatites					Metasomatic rock	
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	
SiO <sub>2</sub>	69.36	71.44	73.69	71.39	73.11	69.56	74.66	71.96	72.88	73.30	74.83	73.39	74.34	74.41	70.73	70.68	60.48
TiO <sub>2</sub>	0.41	0.38	0.23	0.09	0.11	—	0.05	0.28	0.28	0.17	0.13	—	—	0.03	—	—	0.25
Al <sub>2</sub> O <sub>3</sub>	14.73	14.32	13.86	15.49	15.11	18.38	14.45	13.89	13.70	14.12	13.82	15.71	15.17	15.54	17.68	17.32	22.09
Fe <sub>2</sub> O <sub>3</sub>	0.51	0.46	0.32	0.78	0.74	0.15	0.32	0.40	0.30	0.32	0.38	0.24	0.12	0.13	0.12	0.08	0.26
FeO	1.77	1.71	1.31	0.51	0.21	0.18	0.13	1.88	1.57	1.13	0.89	0.56	0.63	—	—	—	0.14
MnO	0.03	0.03	0.03	0.02	0.02	0.04	0.02	0.03	0.04	0.03	0.03	0.12	0.15	0.05	0.12	0.17	—
MgO	0.56	0.50	0.30	0.12	0.28	—	0.08	0.50	0.38	0.28	0.20	—	—	—	—	—	0.27
CaO	0.96	0.83	0.59	0.30	0.24	0.04	0.05	1.28	1.03	0.67	0.60	0.11	0.12	0.10	0.11	0.09	0.51
Na <sub>2</sub> O	4.22	2.79	3.01	3.17	4.18	7.17	0.53	3.15	3.36	2.99	3.12	4.34	4.54	4.78	4.84	4.72	3.36
K <sub>2</sub> O	5.43	5.53	4.95	4.13	4.09	1.63	6.81	5.50	4.86	4.81	4.17	3.64	2.63	3.22	3.69	3.77	9.74
P <sub>2</sub> O <sub>5</sub>	0.32	0.30	0.29	0.37	0.51	0.15	0.17	0.21	0.19	0.27	0.28	0.13	0.19	0.21	0.27	0.18	0.51
F	0.12	0.12	0.13	0.09	0.12	0.14	0.14	0.15	0.16	0.18	0.19	0.42	0.42	0.61	1.51	1.55	0.18
LOI	1.47	1.34	1.24	2.53	0.64	1.64	2.24	1.11	1.26	1.36	1.24	1.17	1.18	0.75	1.20	1.03	1.76
	99.89	99.75	99.97	98.99	99.36	99.08	99.65	100.34	100.01	99.63	99.88	99.83	99.49	99.83	100.27	99.59	99.55
O/F	0.05	0.05	0.06	0.04	0.05	0.06	0.06	0.06	0.06	0.07	0.08	0.08	0.18	0.18	0.26	0.63	0.65
Total	99.84	99.70	99.91	98.95	99.31	99.02	99.59	100.28	99.94	99.55	99.80	99.65	99.31	99.57	99.64	98.94	99.47
A/CNK	1.01	1.18	1.22	1.52	1.27	1.28	1.73	1.03	1.08	1.25	1.29	1.39	1.44	1.35	1.45	1.44	—
Nb	13	12	11	21	12	13	10	18	18	17	19	30	38	29	25	61	8
Ta	*	*	*	*	*	*	*	*	*	*	*	7	13	33	26	62	*
Zn	64	68	45	44	43	36	40	77	73	66	58	116	133	17	31	49	20
Sn	12	14	22	25	23	33	35	19	25	29	45	77	114	224	238	404	54
Li	108	104	120	122	122	163	170	179	195	260	320	743	790	1258	3345	3856	158
Zr	173	164	99	31	24	7	29	161	134	89	65	13	66	10	20	34	99
Cu	15	13	*	34	4	7	8	15	17	*	*	*	*	*	*	4	6
Y	17	15	10	8	*	*	*	27	26	20	16	*	*	*	*	*	5
Sr	106	102	65	19	10	7	8	72	65	63	38	5	4	14	14	12	10
Pb	35	55	23	30	20	24	115	34	22	52	31	13	23	33	23	26	50
Ba	490	399	210	43	18	21	17	320	297	212	121	16	18	31	27	25	440
Rb	246	250	258	269	274	316	314	331	340	386	480	853	784	929	1604	1803	458
Th	20	20	11	*	*	*	*	32	21	14	10	*	*	*	*	*	5

LEBCS - less evolved beryl-columbite, MEBCS – more evolved beryl-columbite, LS - lepidolite; — not detected; \* - below the detection limit. The A/CNK ratio is the molecular Al<sub>2</sub>O<sub>3</sub>/(CaO + Na<sub>2</sub>O + K<sub>2</sub>O), which shows that the granitic rocks are peraluminous.



**Fig. 4.** Variation diagrams of granitic rocks from the Gravanho-Gouveia area, central Portugal define two series: a) granite G1, less evolved beryl-columbite pegmatites and more evolved beryl-columbite pegmatites; b) granites G2 and G3 and lepidolite pegmatites.



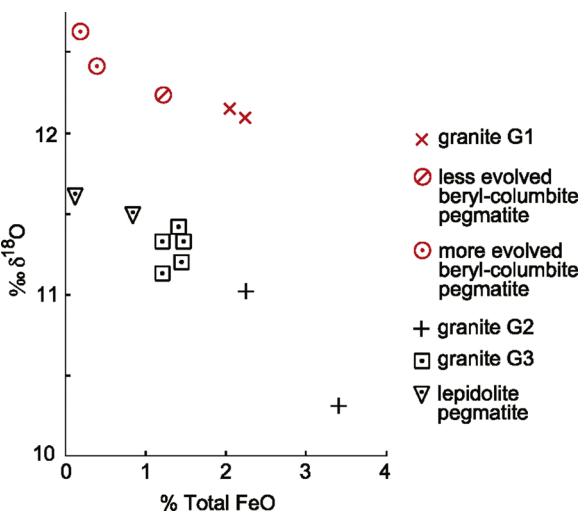
**Fig. 5.** Chondrite-normalized REE abundances in granitic rocks from the Gravanho-Gouveia area, central Portugal. Granite G1, less evolved beryl-columbite pegmatites and more evolved beryl-columbite pegmatites define subparallel REE patterns. Granites G2 and G3 and lepidolite pegmatites define subparallel REE patterns.

**Table 5**

Representative rare-earth element concentrations (ppm) of granites and granitic aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

Granite	LEBCS		MEBCS		Granites		LS pegmatites	
	G1	pegmatites	G2	G3	13	14		
La	28.40	4.12	2.16	36.00	19.45	2.46	1.50	
Ce	59.25	8.31	4.82	78.55	36.25	3.48	3.20	
Pr	7.17	1.02	0.57	9.56	4.41	0.42	0.40	
Nd	26.35	4.08	2.00	34.00	15.52	1.67	1.32	
Sm	5.15	1.17	0.59	7.35	3.58	0.42	0.31	
Eu	0.79	0.22	0.10	0.58	0.29	0.11	0.07	
Gd	4.35	0.93	0.51	6.35	3.18	0.44	0.39	
Tb	0.68	0.14	0.09	0.98	0.54	0.07	0.06	
Dy	3.38	0.65	0.49	5.16	2.72	0.29	0.26	
Ho	0.62	0.07	0.06	1.02	0.51	< 0.06	< 0.06	
Er	1.58	0.17	0.15	2.69	1.28	< 0.06	< 0.06	
Tm	0.25	< 0.04	< 0.04	0.39	0.20	< 0.04	< 0.04	
Yb	1.76	0.18	0.17	2.50	1.25	< 0.03	< 0.04	
Lu	0.26	< 0.02	< 0.02	0.40	0.18	< 0.02	< 0.02	

LEBCS - less evolved beryl-columbite, MEBCS - more evolved beryl-columbite, LS - lepidolite.



**Fig. 6.** Diagram of  $\delta^{18}\text{O}$  versus total FeO of granites G1, G2 and G3 and granitic pegmatites from the Gravanho-Gouveia area, central Portugal, define two distinct trends, one for granite G1, less evolved beryl-columbite pegmatites and another for granites G2 and G3 and lepidolite pegmatites.

**Table 6**  
 $\delta^{18}\text{O}$  of selected samples of granites and granitic aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

	$\delta^{18}\text{O}$
Granite G1	12.1
	12.1
LEBCS	12.2
MEBCS	12.4
pegmatites	12.6
Granite G2	10.3
	11.0
Granite G3	11.1
	11.2
	11.2
	11.3
	11.4
LS pegmatites	11.5
	11.6

LEBCS - less evolved beryl-columbite, MEBCS - more evolved beryl-columbite, LS - lepidolite.

## 6. Geochemistry of minerals

### 6.1. Feldspars

Compositions of feldspars from granites and aplite-pegmatites from the Gravanho-Gouveia area are given in Table 7. The anorthite content of plagioclase decreases progressively from the granite G1 to the less and more evolved beryl-columbite pegmatites, and from granite G2 to granite G3 and lepidolite pegmatites. The K-feldspar from these pegmatites tends to have higher orthoclase content and lower BaO content than the K-feldspar from granite G1. The BaO content of K-feldspar phenocrysts from lepidolite pegmatites is lower than those of K-feldspar phenocrysts from granites G2 and G3.

In the three granite types, there is no significant difference in the contents of  $\text{P}_2\text{O}_5$  in feldspar phenocryst and matrix. The  $\text{P}_2\text{O}_5$  content in K-feldspar and plagioclase from pegmatites is higher than in feldspars from the respective granites (Table 8). The  $\text{P}_2\text{O}_5$  content of plagioclase is not related to the anorthite content of plagioclase. The content of  $\text{P}_2\text{O}_5$  in K-feldspar is higher than that of coexisting plagioclase (Table 8), which agrees with the findings of London et al. (1990); Neiva et al. (2008, 2012) and Antunes et al. (2013), for example. The distribution coefficient of P between K-feldspar and plagioclase pairs ( $D[\text{P}] \text{Kf/Pl}$ ) increases from granite G1 to less and more evolved beryl-columbite pegmatites (Table 8),

### 6.2. Micas

The properties from granitic aplite-pegmatites from the Gravanho-

**Table 7**

Compositions of feldspars from granites and granitic aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

	Granite G1	LEBCS pegmatites	MEBCS pegmatites	Granite G2	Granite G3	LS pegmatites
Anorthite content of plagioclase	Phenocryst	5-30	—	18-29	1-16	—
	Matrix	2-24	3-8	1-16	0-11	0-2
Orthoclase content of K-feldspar	Phenocryst	90-94	—	90-100	93-100	—
	Matrix	88-97	91-98	87-100	94-100	94-98
wt.% BaO of K-feldspar	Phenocryst	$\leq 0.06-0.29$	—	$\leq 0.06-0.18$	$\leq 0.06-0.08$	—
	Matrix	$\leq 0.06-0.14$	$\leq 0.06$	$\leq 0.06$	$\leq 0.06$	$\leq 0.06$

LEBCS - less evolved beryl-columbite, MEBCS - more evolved beryl-columbite; LS - lepidolite; — not found.

**Table 8**

Phosphorus content of feldspars from granites and granitic aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

$\text{P}_2\text{O}_5$ wt.%		Mean	Min/Max	n	$D[\text{P}] \text{Kf/Pl}$
Granite	Kf	0.13	0/0.27	18	1.86
G1	Pl	0.07	0/0.11	18	
LEBCS	Kf	0.44	0.36/0.59	13	1.91
pegmatite	Pl	0.23	0.10/0.44	12	
MEBCS	Kf	1.33	0.88/1.68	18	2.33
pegmatite	Pl	0.57	0.14/0.73	15	
Granite	Kf	0.12	0/0.20	18	1.50
G2	Pl	0.08	0/0.12	19	
Granite	Kf	0.19	0/0.35	10	1.58
G3	Pl	0.12	0/0.21	19	
LS	Kf	0.35	0.01/0.55	21	1.46
pegmatite	Pl	0.24	0/0.36	31	

LEBCS - less evolved beryl-columbite, MEBCS - more evolved beryl-columbite, LS - lepidolite.

Gouveia area are shown in Table 2. The chemical compositions of micas from granites and granitic aplite-pegmatites from the Gravanho-Gouveia area are given in Table 9. The formulae were recalculated to  $22^+$ -cation charges to be plotted in the diagram of Tischendorf et al. (2004) (Fig. 7a). Granite samples of G1, G2 and G3 and less and more evolved beryl-columbite pegmatites contain magmatic muscovite (Fig. 7b). Lepidolite pegmatites contain magmatic muscovite, Li-mica apparently ranging from zinnwaldite to lepidolite (ca. 0.9 to 1.8 apfu Li, Table 9) and late nearly stoichiometric muscovite (Fig. 7b).

Magmatic muscovites from granite G1 and the less and more evolved beryl-columbite pegmatites have progressively higher contents of  $\text{Al}^{\text{VI}}$ , Li, Na and F and lower contents of Ti,  $\text{Fe}^{2+}$  and Mg (Table 9). These muscovites delineate a trend in the  $\text{Al}^{\text{VI}}$  versus  $\text{Fe}^{2+}$  diagram (Fig. 8). Similarly, the contents of Si,  $\text{Al}^{\text{VI}}$ , Li and F increase progressively and Ti,  $\text{Fe}^{2+}$ , Mg and OH contents decrease progressively, going from the magmatic muscovite of the granite G2 to muscovites from granite G3 and lepidolite pegmatites (Table 9). The trend established with these muscovites is parallel and close to the trend defined by muscovites of granite G1 and beryl-columbite pegmatites (Fig. 8). The late muscovite that partially surrounds trilithionite (Fig. 3d) contains more  $\text{Al}^{\text{VI}}$  and K and less Si, Ti,  $\text{Fe}^{2+}$ , Mn, Mg, Li, Na and F than the magmatic muscovite of lepidolite pegmatites (Table 9).

### 6.3. Other silicate minerals

The composition of topaz from lepidolite aplite-pegmatites from the Gravanho-Gouveia area is given in Table 10. It contains 18.9 wt.% F which is in the range of topaz from the S-type and A-type granites in the Krušné Hory/Erzgebirge (17.9–19.8 wt.% F) (Breiter et al., 2013) and more than in topaz from lepidolite aplite-pegmatite sills (5.7 wt.% F) from the Guarda-Belmonte area, in central Portugal (Neiva and Ramos, 2010).

**Table 9**

Representative chemical analyses in wt.% of primary micas and a late muscovite from granites and aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

wt.%	Granite	LEBCS	MEBCS	Granites		LS						
				pegmatites		G2	G3	pegmatites				
				Mu	Mu			Mu	Zin	Tri <sup>+</sup>	Pol <sup>+</sup>	Late Mu
SiO <sub>2</sub>	46.84	47.43	46.66	46.31	46.48	47.06	45.36	51.61	57.48	45.90		
TiO <sub>2</sub>	0.61	—	—	1.01	0.51	0.10	0.61	0.04	0.04	—		
Al <sub>2</sub> O <sub>3</sub>	35.30	35.15	36.61	34.20	34.80	35.51	23.52	25.52	17.92	37.70		
FeO	1.24	0.79	0.38	2.10	1.70	1.20	9.66	0.18	0.18	0.08		
MnO	—	0.02	0.01	—	—	0.25	0.25	1.24	0.48	0.09		
MgO	0.51	0.08	0.09	0.95	0.59	0.11	2.09	—	0.02	—		
Li <sub>2</sub> O*	0.07	0.10	0.14	0.34	0.38	0.45	3.45	5.26	6.95	0.06		
CaO	0.03	0.03	0.01	0.03	0.08	0.01	0.03	0.02	0.02	—		
Na <sub>2</sub> O	0.02	0.14	0.33	—	0.02	0.37	0.13	0.22	—	0.05		
K <sub>2</sub> O	10.90	11.32	10.96	10.80	10.55	10.35	10.42	10.50	10.68	11.09		
Cl	—	—	—	—	—	—	—	—	0.01	0.01		
F	0.26	0.35	0.45	0.90	0.98	1.10	2.80	7.13	9.22	0.24		
H <sub>2</sub> O*	4.40	4.34	4.31	4.08	4.04	4.01	2.99	1.18	0.21	4.41		
	100.18	99.75	99.95	100.72	100.13	100.52	101.31	103.20	103.71	99.63		
O≡F	0.11	0.15	0.19	0.38	0.41	0.46	1.18	2.99	3.87	0.10		
Total	100.07	99.60	99.76	100.34	99.72	100.06	100.13	100.21	99.84	99.53		
Si	3.105	3.155	3.090	3.080	3.096	3.113	3.151	3.395	3.758	3.045		
Al <sup>IV</sup>	0.895	0.845	0.910	0.920	0.904	0.887	0.849	0.605	0.242	0.955		
ΣT	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00		
Al <sup>VI</sup>	1.863	1.910	1.948	1.761	1.827	1.881	1.076	1.374	1.139	1.992		
Ti	0.030	—	—	0.051	0.026	0.005	0.032	0.002	0.002	—		
Fe <sup>2+</sup>	0.069	0.044	0.021	0.117	0.095	0.066	0.561	0.010	0.010	0.004		
Mn	—	0.001	0.001	—	—	0.014	0.015	0.069	0.027	0.005		
Mg	0.050	0.008	0.009	0.094	0.059	0.011	0.216	—	0.002	—		
Li	0.018	0.026	0.036	0.092	0.103	0.119	0.964	1.391	1.828	0.016		
Σ	2.03	1.99	2.02	2.12	2.11	2.10	2.86	2.85	3.01	2.02		
Ca	0.002	0.002	0.001	0.002	0.006	0.001	0.002	0.001	0.001	—		
Na	0.003	0.018	0.042	—	0.003	0.047	0.018	0.028	—	0.006		
K	0.922	0.961	0.926	0.916	0.896	0.873	0.923	0.881	0.891	0.938		
Σ	0.93	0.98	0.97	0.92	0.91	0.92	0.94	0.92	0.91	0.94		
Cl	—	—	—	—	—	—	—	—	0.001	—		
F	0.055	0.074	0.094	0.189	0.206	0.230	0.615	1.484	1.907	0.050		
OH <sup>*</sup>	1.945	1.926	1.906	1.811	1.794	1.770	1.385	0.516	0.092	1.950		

LEBCS - less evolved beryl-columbite, MEBCS - more evolved beryl-columbite, LS - lepidolite; Mu - muscovite, Zin - zinnwaldite, Tri - trilithionite, Pol - poly-lithionite; trilithionite and polylithionite correspond to the lepidolite series; — below the detection limit; Tri<sup>+</sup> contains Rb<sub>2</sub>O 0.30 and 0.013 pfu of Rb; Pol<sup>+</sup> contains 0.50 wt.% Rb<sub>2</sub>O and 0.021 pfu of Rb; Li<sub>2</sub>O\* values calculated from the equations of Tischendorff et al. (1997); OH<sup>\*</sup> calculated by difference to 2.000; H<sub>2</sub>O\* calculated by stoichiometry. Number of ions on the basis of 12 (O, OH, F).

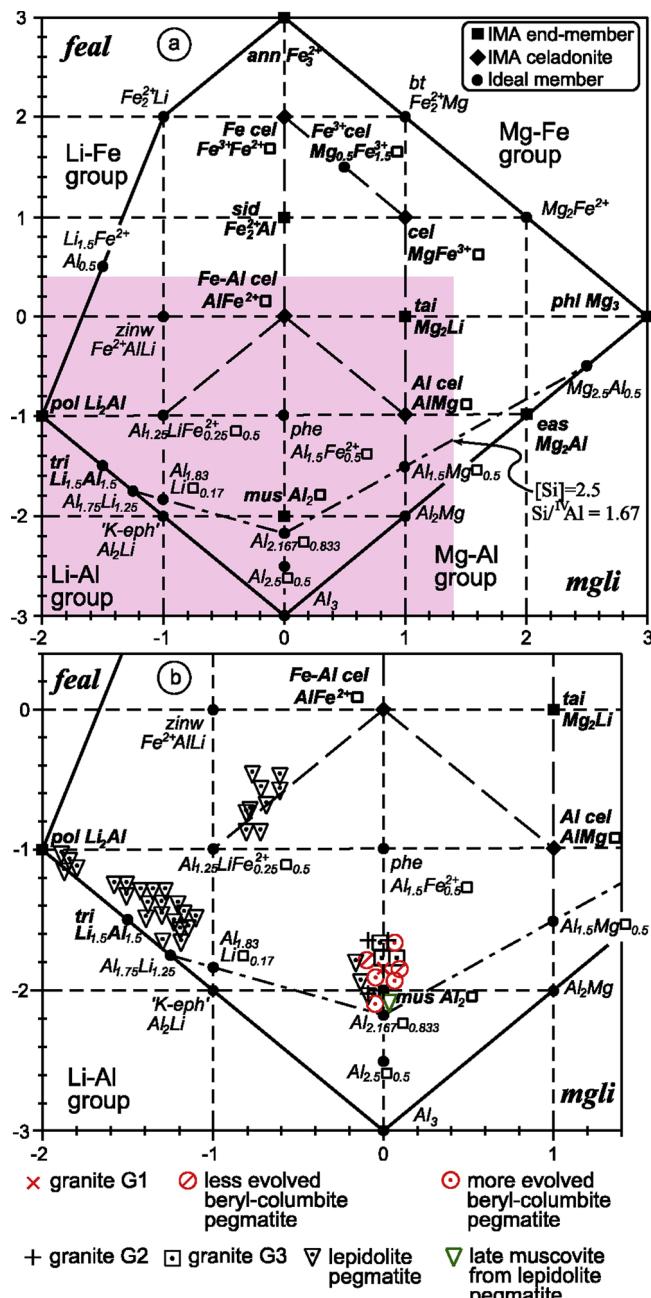
Selected compositions of tourmaline from granitic aplite-pegmatites from the Gravanho-Gouveia area are given in Table 10. Unzoned tourmaline crystals occur in beryl-columbite pegmatites and lepidolite pegmatites, but zoned crystals only occur in the former pegmatites. Compositions of unzoned crystals and rims of zoned crystals belong to the Na(+K) group, whereas the cores of zoned crystals belong to the vacancy group (Fig. 9a, b). The former are schorl and the latter are foitite compositions (Fig. 9c). The change in composition from foitite core to schorl rim defines a discontinuous chemical zoning with a decrease in <sup>Y</sup>Mg/<sup>Y</sup>(Mg + Fe<sup>2+</sup>), X-vacancy/(Na + X-vacancy) and an increase in <sup>Y</sup>Fe<sup>2+</sup>, Na and F contents (Fig. 9d, e, Table 10), probably formed during closed-system growth. This also occurs in some zoned magmatic tourmaline crystals from granitic pegmatites from Jales-Gralheira, in northern Portugal (Neiva et al., 2007). Schorl from lepidolite pegmatites has higher Ti, Ca and F contents and lower <sup>Y</sup>Al content than schorl from beryl-columbite pegmatites and plots outside the trends defined by the core and rim of zoned crystals from beryl-columbite pegmatites (Table 10, Fig. 9).

In the Gravanho-Gouveia area, beryl crystals occur in more evolved beryl-columbite pegmatites (Table 10). The contents of Ti, Mn, Mg, Ca and K are low. Lithium substitutes for Be in the tetrahedral site with the charge balance by alkali ions of suitable radii into the channels (Aurisicchio et al., 1988), particularly Na<sup>+</sup> and Cs<sup>+</sup> (Hawthorne and Huminicki, 2002; Černý et al., 2003; Neiva and Neiva, 2005).

#### 6.4. Oxide minerals

Some chemical analyses of cassiterite from lepidolite pegmatites of the Gravanho-Gouveia area are given in Table 11. Some crystals have slightly more Nb than Ta, whereas others have more Ta than Nb (Fig. 10a), as in cassiterite from the tin-enriched granitic pegmatites from northern Portugal (Neiva, 1996). In zoned crystals, the lighter zone in backscattered images has higher contents of Ta, Nb and Mn and lower content of Sn than the darker zone (Fig. 10a, Table 11). The unzoned crystals and the lighter zone of other crystals have higher Mn content than Fe content (Table 11), which is unusual, but also occurs in cassiterite crystals from lepidolite aplite-pegmatite veins from Segura, central Portugal (Antunes et al., 2013). The compositions of cassiterite plot close to the trend defined by the ideal columbite substitution (Fe, Mn)<sup>2+</sup> + 2(Nb, Ta)<sup>5+</sup> ⇌ 3(Sn, Ti)<sup>4+</sup> (Černý et al., 1985b) (Fig. 10c).

Rutile from lepidolite pegmatites occurs as homogeneous or heterogeneous grains (Fig. 10d, Table 11): the latter have a lighter zone with higher Nb and Fe<sup>2+</sup> contents, similar to higher Ta content and lower Ti content than the darker zone. The lighter zone contains up to 3.4 wt.% FeO, 10.9 wt.% Ta<sub>2</sub>O<sub>5</sub> and 5.5 wt.% Nb<sub>2</sub>O<sub>5</sub>. There is a progressive increase in Nb + Ta and a progressive decrease in Sn + Ti + W from the homogeneous grains to the darker zone and lighter zone of heterogeneous grains (Fig. 10d). The rutile compositions plot close to

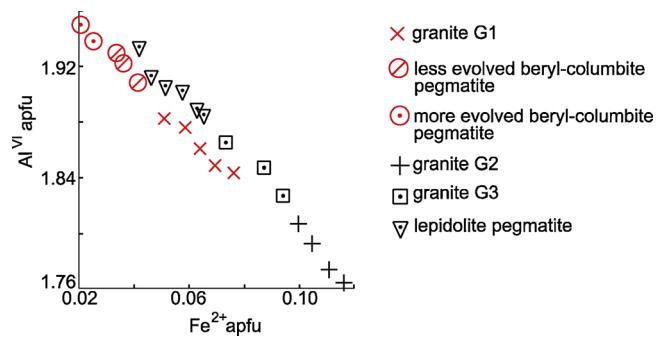


**Fig. 7.** Plots of micas from granites and granitic pegmatites from the Gravanho-Gouveia area, central Portugal, on the diagram for potassium micas of Tischendorf et al. (2004). a. the diagram of feal ( $\text{Fe} + \text{Mn} + \text{Ti-Al}^{\text{VI}}$ ) vs. mgli ( $\text{Mg-Li}$ ) of Tischendorf et al. showing the area of the diagram b. Al cel – aluminoceladonite, ann – annite, bt – biotite, cel – celadonite, eas – eastonite, Fe-Al cel – ferro-aluminoceladonite, Fe cel – ferroceldalomite,  $\text{Fe}^{3+}$  cel – ferriceldalomite, K-eph – K-ephesite, mus – muscovite, phe – phengite, phl – phlogopite, pol – polyliithionite, sid – siderophyllite, tai – tainiolite, tri – trilithionite, zinw – zinnwaldite. b. Compositions of analysed micas from the Gravanho-Gouveia area.

the  $(\text{Fe}, \text{Mn}) (\text{Nb}, \text{Ta})_2$  join.

## 7. Discussion

Based on the geochemistry of major and trace elements, REE, Sr and Sm-Nd isotopic data,  $\delta^{18}\text{O}$  values of granites and the geochemistry of their minerals, the biotite > muscovite granite G1 and muscovite >



**Fig. 8.** The  $\text{Al}^{\text{VI}}$  vs.  $\text{Fe}^{2+}$  diagram of magmatic muscovites from granites and granitic pegmatites from the Gravanho-Gouveia area, central Portugal. Two close distinct trends, one for muscovites from granite G1, less evolved beryl-columbite pegmatites and more evolved beryl-columbite pegmatites and another for muscovites from granites G2 and G3 and lepidolite pegmatites.

biotite granodiorite to granite G2 correspond to two distinct pulses of granite magma derived by partial melting of mid-crustal metasediments, mainly metapelitic and some metagraywacke (Neiva et al., 2009). The muscovite > biotite granite G3 is derived from the granodiorite magma G2 by the fractional crystallization of plagioclase, biotite, quartz and ilmenite (Neiva et al., 2009).

The less and more evolved beryl-columbite pegmatites are paragenetically similar. These pegmatites show a gradational transition to the host granite G1 (Fig. 2); variation diagrams for oxides and trace elements (Fig. 4) and the subparallel REE patterns (Fig. 5) suggest that they constitute a single evolutionary series. This trend is further corroborated with the  $\delta^{18}\text{O}$  versus total FeO diagram (Fig. 6). Granites G2 and G3 and the lepidolite pegmatites define another fractionation trend, which suggests a series distinct from that for the host granite G1 and associated pegmatites (Fig. 4). The subparallel REE patterns and the  $\delta^{18}\text{O}$  versus total FeO diagram also support the two distinct series (Figs. 5 and 6). The decrease in HREE in each series is due to zircon fractionation (e.g. Mittlefehldt and Miller, 1983), as the Zr content decreases in each series (Table 4). The decreases in MREE and LREE are attributed to fractionation of apatite and monazite, respectively (Bea, 1996). The whole-rock Zr,  $\text{P}_2\text{O}_5$ , La and Ce contents (Tables 4 and 5) are consistent with zircon and monazite saturations and appropriate fractionation of zircon, apatite and monazite. There is an increase in  $\delta^{18}\text{O}$  from granite G1 to more evolved beryl-columbite pegmatites by up to 0.4‰, and up to 0.9‰ from granite G2 to lepidolite pegmatites (Table 6), which is compatible with fractional crystallization processes (e.g., White, 2003). In each series, the anorthite content of plagioclase and BaO content of K-feldspar phenocrysts decrease (Table 7). The  $\text{P}_2\text{O}_5$  contents of K-feldspar and plagioclase of pegmatites are higher than those of feldspars from granites of the respective series (Table 8). Muscovite from each series defines a trend in the  $\text{Al}^{\text{VI}}$  versus  $\text{Fe}^{2+}$  diagram (Fig. 8). The occurrence of topaz and Li-F-rich micas in the lepidolite pegmatites suggest increase of the activity of F in this rock. In the lepidolite pegmatites, the homogeneous crystals of rutile are interpreted as magmatic, whereas the heterogeneous crystals have higher contents of Ta, Nb and  $\text{Fe}^{2+}$  than the homogeneous crystals (Table 11) and are derived from hydrothermal fluids.

The least-squares regression method was used to try to model the evolution of major elements for each series. However, it was impossible to get good results for any series and consequently trace elements cannot be modelled taking into account their evolution. Secondary muscovite replacing mainly plagioclase and hydrothermal columbite-group minerals in all pegmatites (Neiva et al., 2015), the occurrence of chlorite in beryl-columbite pegmatites and some late muscovite surrounding trilithionite, and the heterogenous crystals of rutile in lepidolite pegmatites (Fig. 3d, Table 9; Fig. 3f, Table 11, Fig. 10d) are

**Table 10**

Representative chemical analyses in wt.% of some silicate minerals from granitic aplite-pegmatites from the Gravanho-Gouveia area, central Portugal.

Topaz	Tourmaline				Beryl			
	LS pegmatite	BCS pegmatites						
		Unzoned crystal	Zoned crystals	LS pegmatite	MEBCS pegmatite			
			Core	Rim				
SiO <sub>2</sub>	32.10	SiO <sub>2</sub>	35.75	36.00	34.51	35.60	SiO <sub>2</sub>	66.60
Al <sub>2</sub> O <sub>3</sub>	56.13	TiO <sub>2</sub>	0.43	0.21	0.27	0.86	Al <sub>2</sub> O <sub>3</sub>	18.50
MnO	0.03	B <sub>2</sub> O <sub>3</sub> *	10.44	10.62	10.32	10.44	FeO	0.24
F	18.91	Al <sub>2</sub> O <sub>3</sub>	33.22	34.82	34.95	32.48	MnO	0.01
H <sub>2</sub> O	0.84	FeO	12.27	9.45	13.23	12.00	MgO	0.01
Sum	108.01	MnO	0.08	0.01	0.17	—	TiO <sub>2</sub>	0.01
O=F	7.94	MgO	2.13	2.98	0.83	2.35	CaO	0.01
Total	100.07	CaO	0.07	0.02	0.03	0.44	BeO	13.60
Si	3.928	Li <sub>2</sub> O**	0.58	0.92	0.02	0.70	Li <sub>2</sub> O	0.07
Al	0.072	Na <sub>2</sub> O	1.63	1.52	1.86	1.72	Na <sub>2</sub> O	0.23
Σ	4.00	K <sub>2</sub> O	0.05	0.04	0.02	—	K <sub>2</sub> O	0.01
Al	8.022	H <sub>2</sub> O*	3.41	3.56	3.32	3.32	Rb <sub>2</sub> O	—
Mn	0.003	F	0.40	0.22	0.52	0.59	Cs <sub>2</sub> O	0.07
Σ	8.03	Cl	—	—	—	—	H <sub>2</sub> O <sup>+</sup>	0.51
F	7.317	Sum	100.46	100.37	100.05	100.50	Total	99.87
OH	0.685	OF=F	0.17	0.09	0.22	0.25		
Σ	8.00	Total	100.29	100.28	99.83	100.25	Si	6.000
		T Si	5.949	5.896	5.809	5.932	Be	2.952
		Al	0.051	0.104	0.191	0.068	Li	0.025
		B	3.000	3.000	3.000	3.000	Si	0.018
		Z Al	6.000	6.000	6.000	6.000	Al	—
		Y Al	0.464	0.616	0.743	0.311	Σ	3.00
		Ti	0.054	0.026	0.034	0.108	Al	1.970
		Mg	0.529	0.727	0.208	0.584	Fe <sup>2+</sup>	0.018
		Mn	0.011	0.001	0.024	—	Mn	0.001
		Fe <sup>2+</sup>	1.708	1.294	1.863	1.672	Mg	0.001
		Li	0.388	0.606	0.014	0.469	Σ	1.99
		ΣY	3.15	3.27	2.89	3.14	Ca	0.001
		X Ca	0.012	0.004	0.005	0.078	Na	0.040
		Na	0.526	0.482	0.607	0.557	K	0.001
		K	0.010	0.008	0.004	—	Rb	—
		□	0.452	0.506	0.384	0.365	Cs	0.002
		OH <sup>+</sup> *	3.789	3.886	3.723	3.690	Σ	0.04
		F	0.211	0.114	0.277	0.310	H <sub>2</sub> O	0.307
		Cl	—	—	—	—		
		Name	Schorl	Foite	Schorl	Schorl		

LS - lepidolite, BCS - beryl-columbite, MEBCS - more evolved beryl-columbite. Atomic contents for topaz normalized to 24 oxygen atoms and H<sub>2</sub>O estimated assuming that F + OH = 8.00. Structural formulae are based on 31 anions (O, OH, F) for tourmaline and 18 oxygen atoms for beryl. \*\* calculated according to the method of Pesquera et al. (2016). \* Amount inferred from considerations of stoichiometry. — Not detected. Core - darker zone and Rim-lighter zone (Figs. 3a and 9).

findings that suggest that they present some metasomatic effects due to hydrothermal fluids and can explain why it is difficult to model major elements.

The hydrothermal alteration of granites, particularly pegmatites, documents the presence of hydrothermal fluids, which certainly affected most of the granitic rocks studied. However, based on variations in the major and trace elements of apparently unaltered granitic rocks and variations in their mineral compositions, it is possible to conclude that aplite-pegmatites from the Gravanho-Gouveia area are derived from their parent granite magma in each series by fractional crystallization of quartz, plagioclase, K-feldspar, biotite and ilmenite.

The more evolved beryl-columbite pegmatites were hydrothermally altered into episyenite, with gains in Al<sub>2</sub>O<sub>3</sub>, K<sub>2</sub>O and Rb (Table 4) due to microclinization and elevated F and Sn contents attributed to some fluorite and cassiterite, respectively, and loss of SiO<sub>2</sub> due to a decrease in quartz and Nb and Zn contents, which is attributed to the alteration of muscovite.

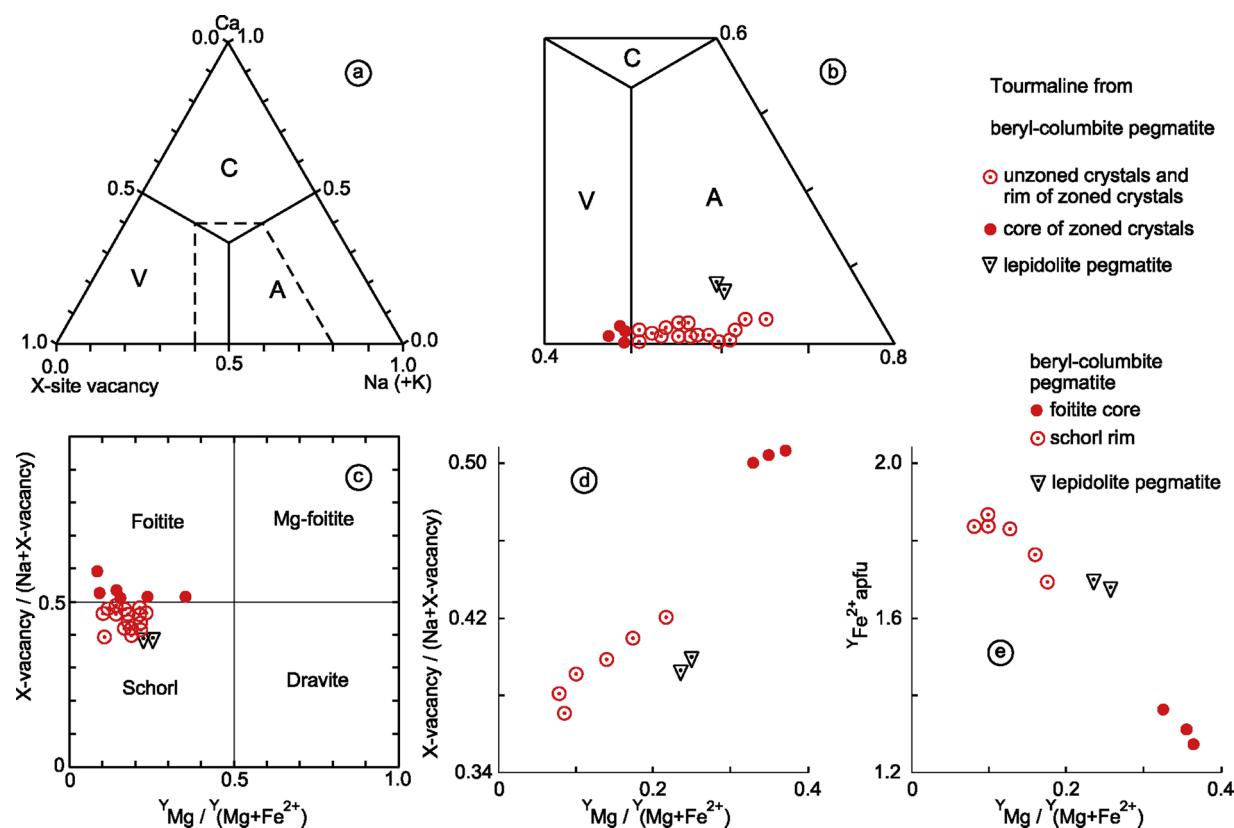
## 8. Conclusions

Field observations together with the petrography, mineralogy and

geochemistry of granites and associated aplite-pegmatite dykes and sills in the Gravanho-Gouveia area indicate the presence of two series of granitic rocks. In each series, pegmatites evolved via the fractional crystallization of quartz, plagioclase, K-feldspar, biotite and ilmenite from the respective parent granitic magma. Cassiterite occurs in lepidolite pegmatites.

Chemically distinct hydrothermal fluids caused the crystallization of a second generation of columbite-tantalite in all pegmatites (Neiva et al., 2015). Another hydrothermal fluid caused the crystallization of secondary muscovite and chlorite in the beryl-columbite pegmatites. Other hydrothermal fluids caused the crystallization of a late muscovite and heterogeneous zoned crystals of rutile in the lepidolite pegmatites.

The most hydrothermally altered rocks are episyenites, which are reddish in colour. The major and trace elements of hydrothermally altered rocks suggest that they are derived from the hydrothermal alteration of a more evolved beryl-columbite pegmatite due to regional circulation of fluids in the granitic rocks.



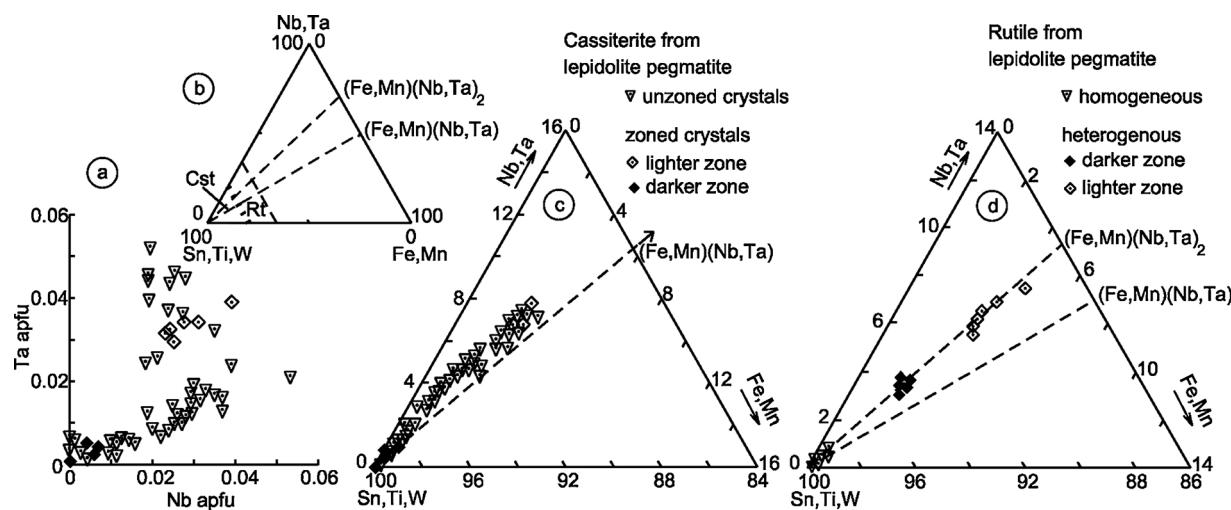
**Fig. 9.** Tourmaline compositions from granitic pegmatites from the Gravanho-Gouveia area, central Portugal. a. Classification of the principal tourmaline groups based on the X-site occupancy (Henry et al., 2011), showing the location of diagram b, C – calcic group, V – vacancy group, A – alkali group. b. unzoned crystals and rims of other crystals plot in the Na(+K) group, but cores plot in the vacancy group. c. X-vacancy/(Na + X-vacancy) versus  $\text{Y}_{\text{Mg}} / \text{Y}(\text{Mg} + \text{Fe}^{2+})$  diagram distinguishing foitite cores from schorl rims and unzoned schorl crystals. d, e. Compositional variation in zoned crystals from beryl-columbite pegmatites and unzoned crystals from lepidolite pegmatites.

**Table 11**

Selected chemical data in wt.% of cassiterite and rutile from lepidolite pegmatites from the Gravanho-Gouveia area, central Portugal.

	Cassiterite				Rutile			
	u	d, l	d	l	Homogeneous		Heterogeneous	
					Darker zone	Lighter Zone	Darker zone	Lighter Zone
WO <sub>3</sub>	—	0.18	—	0.44	WO <sub>3</sub>	0.16	0.21	1.77
Ta <sub>2</sub> O <sub>5</sub>	6.74	1.27	0.38	3.70	Ta <sub>2</sub> O <sub>5</sub>	0.04	6.58	6.36
Nb <sub>2</sub> O <sub>5</sub>	1.66	1.03	0.16	1.97	Nb <sub>2</sub> O <sub>5</sub>	0.37	1.97	4.89
SnO <sub>2</sub>	90.17	97.32	99.74	92.84	SnO <sub>2</sub>	0.58	1.12	5.50
TiO <sub>2</sub>	0.11	0.04	0.01	0.10	TiO <sub>2</sub>	98.03	87.97	1.46
Bi <sub>2</sub> O <sub>3</sub>	—	0.02	0.06	0.04	Bi <sub>2</sub> O <sub>3</sub>	—	0.05	1.28
FeO	0.13	0.03	—	0.06	Fe <sub>2</sub> O <sub>3</sub>	0.40	0.75	0.46
MnO	1.04	0.17	0.01	0.70	FeO	—	1.36	0.18
Total	99.85	100.06	100.36	99.85	MnO	—	0.02	2.73
					Total	99.58	100.03	3.41
W	—	0.001	—	0.003				0.07
Ta	0.046	0.009	0.003	0.025	W	0.001	0.001	0.007
Nb	0.019	0.012	0.002	0.022	Ta	—	0.025	0.007
Sn	0.904	0.970	0.994	0.926	Nb	0.002	0.013	0.037
Ti	0.002	0.001	—	0.002	Sn	0.003	0.006	0.032
Bi	—	—	—	—	Ti	0.990	0.931	0.008
Fe	0.003	0.001	—	0.001	Bi	—	—	0.008
Mn	0.022	0.004	—	0.015	Fe <sup>3+</sup>	0.004	0.008	0.006
Total	0.996	0.998	0.999	0.994	Fe <sup>2+</sup>	—	0.016	0.005
					Mn	—	—	0.002
					Total	1.000	1.000	0.001
								1.000

u - unzoned, d, l - average of darker and lighter zone, d - darker zone, l - lighter zone in backscattered images; — not detected. Cation formula of cassiterite and rutile based on 6 and 2 atoms of oxygen, respectively.



**Fig. 10.** Diagrams for cassiterite and rutile from lepidolite pegmatites from the Gravanhão-Gouveia area, central Portugal. a. Ta versus Nb diagram for cassiterite. b. (Nb, Ta) - (Fe, Mn) - (Sn, Ti, W) diagram showing the location of diagrams c for cassiterite (Cst) and d for rutile (Rt). c. The compositions of cassiterite plot close to the trend for the ideal substitution. d. Compositions of rutile plot close to the (Fe, Mn)(Nb, Ta)<sub>2</sub> join.

## Acknowledgments

Thanks are due to M. Machado Leite for most of the analytical data on aplite-pegmatites and their minerals obtained at LNEG; F. Guimarães for helping to get electron microprobe data at LNEG; F. Longstaffe for the determination of oxygen isotope data; M. Rieder for the use of his computer program for micas. We are grateful to Alex Deutsch, Tomas Magna, K. Breiter, E. Roda-Robles and an anonymous reviewer for their helpful suggestions. This research was financially supported by the FCT project UID/GEO/04035/2013.

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