Virtual laboratories in (bio)chemical engineering education

Lucília Domingues*, Isabel Rocha, Fernando Dourado, Madalena Alves, Eugénio C. Ferreira

IBB–Institute for Biotechnology and Bioengineering, Centre of Biological Engineering, Universidade do Minho, Campus de Gualtar, 4710-057 Braga, Portugal

ABSTRACT

In the last decades, Information and Communications Technologies (ICT) have been promoting the creation and adoption of new learning and teaching styles. Virtual laboratories, by overcoming some limitations of conventional hands-on experiments, have been adopted as a complement or in substitution of laboratory sessions.

This paper describes the design and implementation of two virtual labs for biochemical engineering education intended for students at the BSc degree.

One of the virtual labs is intended to fully replace the hands-on experiment and consists on the determination of the correlation between oxygen transfer rate, aeration rate and agitation power in a reactor. The other virtual lab consists on the determination of the residence time distribution (RTD) in continuous stirred tanks series and was implemented to support the physical experiments rather than replacing them.

The virtual labs provide the students a learning platform covering the fundamentals underlying the experiment, its pre-visualization and simulation. The effectiveness of the implemented system was evaluated through direct experimentation and survey (through questionnaires) with students taking the chemical technology lab course. For the RTD virtual Lab, and based on specific learning outcomes, teachers could assess significant improvement in students’ performance in the lab and also a more thorough discussion of the results in the reports. The survey results show that, in average, considering the two virtual labs and several classes, 93% of the students consider the virtual labs of great utility.

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1. Introduction

Information and Communications Technologies (ICT) facilitates the development of novel teaching strategies for laboratory classes, including new approaches to illustration, simulation, demonstration, experimentation, operation, and communication.

While the “hands-on” approach for laboratory experiments has enormous educational value, these traditional teaching methods are expensive and require complex logistics regarding space, staff, scheduling and safety (Feisel and Rosa, 2005; Selmer et al., 2007). Virtual laboratories (herein “virtual labs”) may allow overcoming these limitations by allowing a computerized simulation of the laboratory experiments (Rasteiro et al., 2009). Even though virtual labs cannot fully substitute the hands-on laboratory experiments in engineering curricula, they also provide several advantages as a complementary educational tool, the most notorious being the possibility of performing them anytime at any place provided Internet access is available (Ferreira, 2004). Virtual labs have been considered as a support to physical laboratories (Rafael et al., 2007; Shin et al., 2002) and even remote laboratories may be used as a complement to lab sessions (Klein and Wozny, 2006).

The University of Minho has been promoting a pilot project that concerns the development of components and software agents for assistance in laboratory lessons of curricular units from Science and Technology areas—the V Labs Project. These software contents will lead to the establishment of virtual laboratories and are expected to complement the students’ lab work, as well as to promote new e-Learning education activ-

* Corresponding author. Tel.: +351 253 604400; fax: +351 253 678986.
E-mail address: luciliad@deb.uminho.pt (L. Domingues).
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Fig. 1 – Web platform structure of the “Determination of the Residence Time Distribution in Continuous Tank Series” virtual lab.

2. Implementation and results

2.1. Determination of the residence time distribution (RTD) in continuous tank series

2.1.1. Experiment description

This experiment aims at the experimental determination of the residence time distribution in a cascade of 1, 2, 3, and 4 stirred tanks (see for example Fogler, 1986; Levenspiel, 1972 for the fundamentals of this work) and the application of the “Tanks in Series Model” to a series of continuous stirred tank reactors. Students are also encouraged to apply the “Dispersion Model” and to use the Solver feature of the Excel® spreadsheet (Ferreira et al., 2004) to determine the model parameters—the Peclet number (for the dispersion model) and the number of tanks (for the tanks in series model) from the best fits of their experimental data. The experimental RTD is obtained using methylene blue as tracer. After a pulse injection, its concentration profile is followed at the tanks' outlet by spectrophotometry. This experiment has been implemented in the CTL course for many years, as a traditional lab experiment. It was adapted for the VLabs because, even though it is very easy and safe to perform in the laboratory, the students’ written reports generally showed an insufficient knowledge of the RTD concepts, as well as strong limitations in the discussion of the results obtained from their experiments.

2.1.2. Virtual lab goals and gains

The main goal with this virtual lab was to provide a web support to the traditional lab experiment, thus improving student’s effectiveness and autonomy in the laboratory class, in data analysis and report writing.

Even though this virtual lab was not designed to replace the hands-on experiment, it reduced costs as it allowed decreasing the time needed to conduct the experiment, decreasing the number of repetitions (due to errors), and also eliminating the need for full-time supervision.

2.1.3. Web platform description

The web platform structure is presented in Fig. 1 and is divided into three main sections: theory fundamentals, experiment and simulation.

In the theory fundamentals section, the theory underlying the experiment is presented with animations and illustrations. Also, a self-evaluation test can be done and some exercises are proposed.

In the experiment section, the safety procedures that students must learn and practice once in the lab are given. Also, a detailed description of the experimental set-up is provided through an interactive platform, where the students can view details on specific equipment just by placing the computer mouse on top of the corresponding picture. To enhance the visual experience, parts of the experimental procedure are detailed in videos that show some relevant aspects of the experimental assay, accompanied by written text and sound, explaining what is being executed. In this section, the experimental protocol and a template for lab data registration are also available for download, so that students can take it to the
lab. Also, there is a calculation procedure implemented in a questionnaire format so that the students can think by themselves before the answer is provided by the software. Finally, a test-quiz has to be answered by the students. This test is meant to evaluate if the students are prepared to do the laboratory experiment. Once the students complete the test, the results are sent automatically to the teacher by email. If a score above the 50% threshold is obtained, the student is authorized to do the experiment.

In the simulation section the students can perform RTD experiments using either the tanks in series model or the dispersion model. They can set the volumetric flow rate, the tank volume and model parameters (the number of tanks for the tanks in series model and the Peclet number for the dispersion model). Students can also use the simulation section to determine the most suitable time interval for sample collection, a relevant parameter for performing the RTD laboratory experiments.
2.2. Determination of correlations between oxygen transfer rate, aeration rate and agitation power (A&A)

2.2.1. Experiment description
The focus of this virtual lab is on the oxygen transfer from air to liquid media and on the influence of aeration rate and stirring power on the oxygen transfer coefficient (the $k_La$ coefficient) (see for example Geankoplis, 1983; Wang et al., 1979 for the fundamentals of this work). This experiment had been implemented in the CTL course for some years as a traditional lab experiment. It consists in measuring the dissolved oxygen concentration after sparging water in a stirred tank with nitrogen with the subsequent introduction of air. Different aeration rates and stirrer speeds are tested in order to correlate those variables with the oxygen transfer coefficient.

This virtual lab was developed due to the high operational costs with nitrogen, and is intended to completely replace the hands-on experiment. This in fact has other advantages, since the experimental procedure although simple, is repetitive and monotonous.

2.2.2. Virtual lab goals and gains
In summary, the main goal of this implementation was to replace the hands-on experiment, allowing to save time and costs, while improving the students’ education (this includes developing self-study, team work, initiative and organizational skills, to name a few). As a note, before this virtual lab implementation, the laboratory classes would last 6 h. Before the classes, each work-group was required to meet with the teacher for a 15 min oral evaluation and discussion on the protocol, safety and waste disposal procedures. Now, 20 min are more than adequate to perform the simulations using any computer with an Internet connection.

2.2.3. Web platform description
The web platform structure, presented on Fig. 2, is divided into five different sections: theory fundamentals, experimental methodology, calculation methodology, references and laboratory.

The first three sections were designed to support the experiment, results analysis and report writing. In the first section, theory fundamentals are provided while in the second section, a brief description of the aims, procedure and equipment is presented. The detailed experimental protocol is available for download from the procedure section. The third section describes the calculation methodology to be applied to the gathered data.

The fifth section (Laboratory) is the section where the actual virtual experiment can be executed. It is divided into evaluation test and experiment. The students are only authorized to perform the experiment if they obtain more than 50% in an evaluation test. This test is generated automatically by the computer from a pool of questions stored in a database. In the experiment platform, one can choose the air volumetric flow rate and the stirring speed for each experiment. Students have to conduct the same procedure as they would in a conventional lab experiment that is, to first turn on the water supply and fill the tank, and then start the stirrer. They should turn on the nitrogen supply so that the dissolved oxygen concentration drops to zero and then turn on the air supply. Fig. 3 shows the experiment platform. The left hand side is where the students can do the manipulations of the process.
Fig. 5 – Students survey results in aeration and agitation virtual lab (2006/2007 results). Question 1: The A&A virtual Lab contents have led to a better understanding of the concepts necessary for the execution of the experimental work; 2: the virtual experiment replaces the hands-on experiment completely, there is no need to do the experiment on the lab.; 3: the A&A virtual Lab contents have made the calculation procedure easier; 4: the A&A virtual Lab contents have helped in the analysis and interpretation of the results; 5: the A&A virtual Lab portal is easy to use; 6: the A&A virtual Lab portal is well organized; 7: the A&A virtual Lab portal is complete; 8: in general, I consider the A&A virtual Lab portal of great utility.

Data from the survey performed in 2006/2007, 2007/2008 and 2008/2009 for the RTD virtual lab, in a total of 121 students, are presented in Fig. 4.
The students have considered that the virtual lab portal is easy to use (95% agreed or completely agreed) and well organized (93% agreed or completely agreed), and 79% have considered that it is complete. The majority of the students have agreed that the RTD virtual Lab contents led to a better understanding of the concepts necessary for the execution of the experimental work (97%) and that it made easier the execution of the experimental work (94%). They have also considered (88%) that the virtual lab portal helped in the analysis and interpretation of the results. Only 83% have agreed that the RTD virtual lab has made the calculation procedure easier. Overall, 98% of the students have considered the virtual lab portal of great utility.

The students' survey results obtained for the aeration and agitation virtual lab, in a total of 30 students, are presented in Fig. 5.

Again, students have considered that the virtual lab portal is easy to use (97%) and well organized (90%). Interestingly, a higher number of responders disagreed or completely disagreed (44%) when it was stated that the virtual experiment replaced the hands-on experiment completely. Nevertheless, one can see that opinions are quite positive towards the virtual lab portal.

4. Summary and conclusions

This paper describes the design and implementation of two virtual labs for biochemical engineering education. These virtual labs were designed for the course “Chemical Technology Laboratory” taught to the third year students of the Biological Engineering integrated Master studies in Minho University, Portugal. The experiment was first implemented in 2006.

One of the virtual labs consists on the determination of the correlation between oxygen transfer rate, aeration rate and agitation power in a reactor. This virtual lab is intended to fully replace the hands-on experiment. The advantages over performing the experiment are clear: reduced costs, reduced experiment time, and improved data with no loss of education efficiency. The other virtual lab consists on the determination of the residence time distribution (RTD) in continuous tanks series and was implemented to support the experiments rather than replacing them.

These virtual labs demonstrate that, by way of the Internet, there are remarkable tools to support the conventional hands-on laboratory experiments as they provide the students an experimental platform gathering the fundamentals underlying the experiment, its pre-visualization (videos, simulation) and data generation. This web support will improve the students’ effectiveness and autonomy in the laboratory class, as well as in the analysis of the results and report writing.

The effectiveness of the implemented system was evaluated through direct experimentation and survey (through questionnaires) with students taking the chemical technology lab course. Teachers have noticed a clear improvement in students’ performance in the lab and also a more thorough discussion of the results in the reports (for the RTD virtual lab). The survey results show that the majority of the students (93% of a total of 151 students) consider the virtual labs portal of great utility.

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